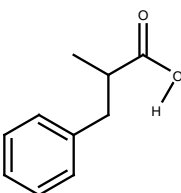
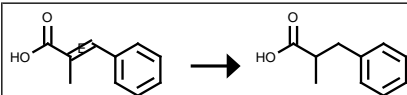
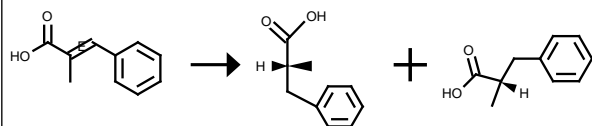


## Query

	Query	Results	Date
1. Query	 <p>Search as: Product, As drawn, Ignore stereo, No salts, No mixtures, No isotopes, No charges, No radicals, No additional rings</p>	188 reactions	2010-06-19 16h:53m:33s (EST)


Rx-ID: 305307 [View in Reaxys](#)

Yield	Conditions & References
99 %	<p><b>With</b> hydrogen, KOH, Pd/polystyrene-poly(ethylene glycol) in H<sub>2</sub>O, Time= 24h, T= 25 °C , p= 760Torr</p> <p><b>Nakao, Ryu; Rhee, Hakjune; Uozumi, Yasuhiro</b>; Organic Letters; <b>vol.</b> 7; nb. 1; (2005); p. 163 - 165  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, &lt;Rh(cycloocten)2Cl&gt;<sub>2</sub>, amidine-Li-der. in tetrahydrofuran, Time= 17h, p= 825.07Torr , Ambient temperature, cat. asymmetric hydrogenation; various times, solvents and mol-ratios of the catalysts; Li-derivative of C<sub>6</sub>H<sub>5</sub>C=N-(S)-CH(CH<sub>3</sub>)(C<sub>6</sub>H<sub>5</sub>)&gt;NHCH(CH<sub>3</sub>)(C<sub>6</sub>H<sub>5</sub>) as cocatalyst, Product distribution</p> <p><b>Brunner, Henri; Agrifoglio, Giuseppe</b>; Monatshefte fuer Chemie; <b>vol.</b> 111; (1980); p. 275 - 287  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> Raney nickel, T= 150 °C , p= 73550.8Torr , Hydrogenation</p> <p><b>Mastagli, Lambert</b>; Comptes Rendus Hebdomadaires des Seances de l'Academie des Sciences; <b>vol.</b> 233; (1951); p. 751  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> nickel-aluminium-alloy, aqueous NaOH</p> <p><b>Schrecker</b>; Journal of Organic Chemistry; <b>vol.</b> 22; (1957); p. 33  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> acetic acid, platinum, Hydrogenation</p> <p><b>Woodruff; Conger</b>; Journal of the American Chemical Society; <b>vol.</b> 60; (1938); p. 465  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, Pd/C</p> <p><b>Somasekharan, K. N.; Kiefer, Edgar F.</b>; Indian Journal of Chemistry, Section B: Organic Chemistry Including Medicinal Chemistry; <b>vol.</b> 27; nb. 1-12; (1988); p. 29 - 37  <a href="#">View in Reaxys</a></p> <p><b>Yu, Hongtao; Simon, Helmut</b>; Tetrahedron; <b>vol.</b> 47; nb. 43; (1991); p. 9035 - 9052  <a href="#">View in Reaxys</a></p>
99 % Chromat.	<p><b>With</b> ammonium formate, [bmim][BF<sub>4</sub>], Pd(OAc)<sub>2</sub>, Time= 5h, T= 65 °C</p> <p><b>Baan, Zoltan; Finta, Zoltan; Keglevich, Gyoergy; Hermecz, Istvan</b>; Tetrahedron Letters; <b>vol.</b> 46; nb. 37; (2005); p. 6203 - 6204  <a href="#">View in Reaxys</a></p>
97 % Spectr.	<p><b>With</b> H<sub>2</sub>, polysilane-supported Pd in hexane, Time= 6h, T= 20 °C</p> <p><b>Oyamada, Hidekazu; Akiyama, Ryo; Hagio, Hiroyuki; Naito, Takeshi; Kobayashi, Shu</b>; Chemical Communications (Cambridge, United Kingdom); nb. 41; (2006); p. 4297 - 4299  <a href="#">View in Reaxys</a></p>
99 % Chromat.	<p><b>With</b> HCO<sub>2</sub>NH<sub>4</sub>, [bmim][BF<sub>4</sub>], palladium/magnesium-lanthanum mixed oxide, Time= 5h, T= 80 °C</p> <p><b>Baan, Zoltan; Potor, Attila; Cwik, Agnieszka; Hell, Zoltan; Keglevich, Gyoergy; Finta, Zoltan; Hermecz, Istvan</b>; Synthetic Communications; <b>vol.</b> 38; nb. 10; (2008); p. 1601 - 1609  <a href="#">View in Reaxys</a></p>


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Yield	Conditions & References
	<p><b>With</b> H<sub>2</sub>, &lt;Rh<sub>2</sub>(COD)<sub>2</sub>&gt;&lt;BF<sub>4</sub>&gt;<sub>2</sub>, Et<sub>3</sub>N <b>in</b> tetrahydrofuran, Time= 1h, p= 760Torr , Ambient temperature, different chiral catalysts, Product distribution</p> <p><b>Johnson, Thomas H.; Rangarajan, G.;</b> Journal of Organic Chemistry; <b>vol.</b> 45; nb. 1; (1980); p. 62 - 65 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, &lt;RhCl(nbd)&gt;<sub>2</sub>, chiral phosphinediamine <b>in</b> methanol, Time= 57h, T= 25 °C , p= 15001.2Torr , various solvents, pressures, times; other chiral phosphine diamine ligand, Product distribution</p> <p><b>Yamada, Issaku; Ohkouchi, Munetaka; Yamaguchi, Motowo; Yamagishi, Takamichi;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 12; (1997); p. 1869 - 1874 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, &lt;RhCl(nbd)&gt;<sub>2</sub>-(S,S)-isopropylbis&lt;2-(1-N,N-dimethylaminoethyl)phenyl&gt;phosphine <b>in</b> methanol, Time= 86h, T= 25 °C , p= 30400Torr , Title compound not separated from byproducts</p> <p><b>Yamada, Issaku; Yamaguchi, Motowo; Yamagishi, Takamichi;</b> Tetrahedron: Asymmetry; <b>vol.</b> 7; nb. 12; (1996); p. 3339 - 3342 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, &lt;RhCl(bicyclo&lt;2.2.1&gt;hepta-2,5-diene)&gt;<sub>2</sub>, (S,S)-isopropylbis&lt;2-(1-N,N-dimethylaminoethyl)phenyl&gt;phosphine <b>in</b> methanol, Time= 86h, T= 25 °C , p= 30002.4Torr , Yield given. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Yamada, Issaku; Ohkouchi, Munetaka; Yamaguchi, Motowo; Yamagishi, Takamichi;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 12; (1997); p. 1869 - 1874 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, Ru(OCOCH<sub>3</sub>)<sub>2</sub>&lt;(S)-2,2'-bis(diphenylphosphino)-5,5',6,6',7,7',8,8'-octahydro-1,1'-binaphthyl&gt; <b>in</b> methanol, Time= 48h, p= 1140Torr , Ambient temperature, Yield given. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Uemura, Toshitsugi; Zhang, Xiaoyoung; Matsumura, Kazuhiko; Sayo, Noboru; Kumobayashi, Hidenori; et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 61; nb. 16; (1996); p. 5510 - 5516 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, [Ru(η<sup>2</sup>-O<sub>2</sub>CCH<sub>3</sub>)<sub>2</sub>((S)-H<sub>8</sub>-binap)] <b>in</b> methanol, Time= 48h, T= 25 °C , p= 1125.09Torr , Title compound not separated from byproducts</p> <p><b>Maienza, Francesca; Santoro, Francesco; Spindler, Felix; Malan, Christophe; Mezzetti, Antonio;</b> Tetrahedron: Asymmetry; <b>vol.</b> 13; nb. 16; (2002); p. 1817 - 1824 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> hydrogen, (R)-(S)-NMe<sub>2</sub>-P(3,5-Me-4-MeOPh)<sub>2</sub>-Mandyphos, [Rh(nbd)<sub>2</sub>]BF<sub>4</sub> <b>in</b> methanol, Time= 94h, T= 25 °C , p= 37503Torr , Title compound not separated from byproducts</p> <p><b>Spindler, Felix; Malan, Christophe; Lotz, Matthias; Kesselgruber, Martin; Pittelkow, Ulrich; Rivas-Nass, Andreas; Briel, Oliver; Blaser, Hans-Ulrich;</b> Tetrahedron: Asymmetry; <b>vol.</b> 15; nb. 14; (2004); p. 2299 - 2306 <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, (R)-1,1'-di[(3,4,5-Me<sub>3</sub>C<sub>6</sub>H<sub>2</sub>)<sub>2</sub>P]-9,9'-spirobifluorene*<b>Ru(OAc)<sub>2</sub></b> <b>in</b> methanol, Time= 18h, T= 25 - 28 °C , p= 4560Torr , Title compound not separated from byproducts</p> <p><b>Cheng, Xu; Zhang, Qi; Xie, Jian-Hua; Wang, Li-Xin; Zhou, Qi-Lin;</b> Angewandte Chemie, International Edition; <b>vol.</b> 44; nb. 7; (2005); p. 1118 - 1121; Angewandte Chemie; <b>vol.</b> 117; nb. 7; (2005); p. 1142 - 1145 <a href="#">View in Reaxys</a></p>

	<p><b>With</b> H<sub>2</sub>, [Rh(cod)2]BF<sub>4</sub>, phosphoramidite based on chiral-3,3'-dimethylbinol, PPh<sub>3</sub> in propan-2-ol, H<sub>2</sub>O, Time= 16h, T= 30 °C , p= 18751.5Torr , Title compound not separated from byproducts</p> <p><b>Hoen, Rob; Boogers, Jeroen A. F.; Bernsmann, Heiko; Minnaard, Adriaan J.; Meetsma, Auke; Tiemersma-Wegman, Theodora D.; Vries, Andre H. M. de; Vries, Johannes G. de; Feringa, Ben L.;</b> Angewandte Chemie, International Edition; <b>vol.</b> 44; nb. 27; (2005); p. 4209 - 4212; Angewandte Chemie; <b>vol.</b> 117; nb. 27; (2005); p. 4281 - 4284  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, (R)-spirofluorene-[(3,4,5-Me<sub>3</sub>-C<sub>6</sub>H<sub>2</sub>)2P]2-[Ru(OAc)2] complex in methanol, Time= 18h, T= 25 - 28 °C , p= 4560Torr , Title compound not separated from byproducts</p> <p><b>Cheng, Xu; Xie, Jian-Hua; Li, Sheng; Zhou, Qi-Lin;</b> Advanced Synthesis and Catalysis; <b>vol.</b> 348; nb. 10-11; (2006); p. 1271 - 1276  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, pyridylphosphorus ligand, zinc(II) salphen, [Rh(nbd)2(BF<sub>4</sub>)] in CH<sub>2</sub>Cl<sub>2</sub>, Time= 16h, T= 35 °C , p= 3750.3Torr , Product distribution, Further Variations: Reagents</p> <p><b>Kuil, Mark; Goudriaan, P. Elsbeth; Kleij, Arjan W.; Tooke, Duncan M.; Spek, Anthony L.; Leeuwen, Piet W. N. M. van; Reek, Joost N. H.;</b> Dalton Transactions; nb. 22; (2007); p. 2311 - 2320  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> [Rh(COD)(S,S)-bis(1-naphthyl(phenyl)phosphino)ethane]]BF<sub>4</sub>, H<sub>2</sub>, Et<sub>3</sub>N in methanol, Time= 65h, T= 25 °C , p= 4500.45Torr , Product distribution, Further Variations: Reagents</p> <p><b>Maienza, Francesca; Spindler, Felix; Thommen, Marc; Pugin, Benoit; Malan, Christophe; Mezzetti, Antonio;</b> Journal of Organic Chemistry; <b>vol.</b> 67; nb. 15; (2002); p. 5239 - 5249  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, Rh/[1-[(1-Ph<sub>2</sub>P)Et]-2-(2-(3,5-(CF<sub>3</sub>)<sub>2</sub>-Ph)2-P-Ph)-ferrocene] in methanol, Time= 20h, T= 25 °C , p= 3750.38Torr , Product distribution, Further Variations: Catalysts</p> <p><b>Sturm, Thomas; Weissensteiner, Walter; Spindler, Felix;</b> Advanced Synthesis and Catalysis; <b>vol.</b> 345; nb. 1-2; (2003); p. 160 - 164  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, 2,6,2',6'-(MeO)4-4,4'-[(3,5-Me<sub>2</sub>-C<sub>6</sub>H<sub>3</sub>)2-P]2-[3,3']bipyridinyl, [Ru(benzene)Cl]Cl in methanol, Time= 24h, T= 60 °C , p= 4500.45Torr , Title compound not separated from byproducts.</p> <p><b>Qiu, Liqin; Li, Yue-Ming; Kwong, Fuk Yee; Yu, Wing-Yiu; Fan, Qing-Hua; Chan, Albert S. C.;</b> Advanced Synthesis and Catalysis; <b>vol.</b> 349; nb. 4-5; (2007); p. 517 - 520  <a href="#">View in Reaxys</a></p>
	<p>Example Name C.12  The catalyst precursor and the ligand were stirred in the solvent under argon. A solution of the substrate was added, the argon is drawn off with vacuum and the vessel is connected to a hydrogen supply at the given pressure and temperature. Switching on the stirrer starts the hydrogenation. After the given time, the stirrer is switched off and the solution is placed under argon again. Conversion and enantiomeric excess (ee) are determined by gas chromatography.</p> <p><b>With</b> hydrogen, di(norbornadiene)rhodium(I) tetrafluoroborate, C<sub>40</sub>H<sub>58</sub>FeN<sub>2</sub>O<sub>3</sub>P<sub>2</sub> in methanol, Time= 40h, T= 25 °C , p= 3750.38Torr , Product distribution / selectivity</p> <p><b>Patent; Solvias AG;</b> WO2007/135179; (2007); (A1) English  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> 5percent-palladium/activated carbon, hydrogen, (L)-(-)-cinchonidine, benzylamine in 1,4-dioxane, water, T= 23 °C , p= 760.051Torr , optical yield given as percent ee, enantioselective reaction</p> <p><b>Sugimura, Takashi; Uchida, Takayuki; Watanabe, Junya; Misaki, Tomonori; Okuyama, Tadashi; Kubota, Takeshi; Okamoto, Yasuaki;</b> Journal of Catalysis; <b>vol.</b> 262; nb. 1; (2009); p. 57 - 64  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> di(norbornadiene)rhodium(I) tetrafluoroborate, hydrogen in tetrahydrofuran, Time= 24h, T= 20 °C , p= 7500.75Torr , optical yield given as percent ee, enantioselective reaction</p>

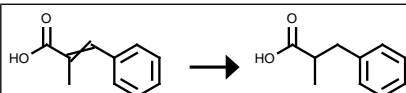
	<p><b>Lee, Jong-Dae; Thanh, Thien Co; Kim, Tae-Jeong; Sang, Ook Kang;</b> Synlett; nb. 5; (2009); p. 771 - 774  <a href="#">View in Reaxys</a></p>
	<p>Example Name D.3  0.41 g (2.53 mmol) of 2-methylcinnamic acid and 5 ml of degassed methanol are introduced in succession into a Schlenk vessel filled with argon. A catalyst solution consisting of 4.73 mg (0.01265 mmol) of [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub>, 8.77 mg (0.0133 mmol) of ligand A and 5 ml of degassed methanol is prepared in a second Schlenk vessel filled with argon. This solution and the catalyst solution are then transferred in succession by means of a steel capillary into a 50 ml autoclave filled with argon. The ratio of substrate/catalyst is 200. The autoclave is closed and a pressure of 5 bar is set by means of 4 flushing cycles (pressurization to 10 bar of hydrogen). The autoclave is thermostated at 25.deg.C and the reaction is started by switching on the stirrer. The reactor is stirred for 19 hours. After opening the autoclave, a reddish reaction solution is isolated. The conversion is quantitative (determined by means of GC and (at)H-NMR). Removal of the solvent on a rotary evaporator gives a quantitative yield of 2-methyl-3-phenylpropionic acid having an enantiomeric purity of 29percent ee (determined by means of HPLC after conversion into the methyl ester; column: Chiracel OB. ).</p> <p><b>With</b> hydrogen, di(norbornadiene)rhodium(I) tetrafluoroborate, (S)Fc-<math>\alpha</math>-S(C<sub>5</sub>H<sub>5</sub>)Fe(C<sub>5</sub>H<sub>3</sub>(PPh<sub>2</sub>)CH(OH)C<sub>6</sub>H<sub>4</sub>(PPh<sub>2</sub>)) in methanol, Time= 19h, T= 25 °C , p= 3750.38Torr , Product distribution / selectivity</p> <p><b>Patent; UMICORE AG and CO. KG;</b> WO2005/108409; (2005); (A2) English  <a href="#">View in Reaxys</a></p>
	<p>A method similar to that of Example D3 is employed. 8.96 mg (0.0133 mmol) of methoxy- TANIAPHOS are used in place of the ligand A. The conversion is 100percent. Removal of the solvent on a rotary evaporator gives a quantitative yield of 2-methyl-3-phenylpropionic acid having an enantiomeric purity of 30percent ee.</p> <p><b>With</b> hydrogen, di(norbornadiene)rhodium(I) tetrafluoroborate, (S)-1-diphenylphosphino-2-[<math>\alpha</math>-(S)-methoxy(o-diphenylphosphinophenyl)methyl]ferrocene in methanol, Time= 19h, T= 25 °C , p= 3750.38Torr , Product distribution / selectivity</p> <p><b>Patent; UMICORE AG and CO. KG;</b> WO2005/108409; (2005); (A2) English  <a href="#">View in Reaxys</a></p>
	<p>Example Name B.1  The method of carrying out the hydrogenations and the determination of the optical yields ee is described in general terms by W. Weissensteiner et al in Organometallics 21 (2002), pages 1766-1774. The catalysts are in each case prepared in situ in the solvent by mixing of the ligand and metal complex as catalyst precursor (unless indicated otherwise = [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub>). Unless indicated otherwise, the substrate concentration is 0.25 mol/l. The determination of conversion and ee of MAA is carried out by means of gas chromatography using a chiral column (Chirasil-L-val).The hydrogenations of EAC are carried out in ethanol in the presence of 5percent (v/v) of CF<sub>3</sub>CH<sub>2</sub>OH. The determination of the ee is carried out by means of gas chromatography using a chiral column [Lipodex E (30m); 1300C isothermal; 190 KPa H<sub>2</sub>]. In the hydrogenation of EOv, [Ru<sub>2</sub>(p-cumene)]<sub>2</sub> is used as metal complex and catalyst precursor. The determination of the ee is carried out after reaction with trifluoroacetic anhydride by means of gas chromatography using a chiral column [Lipodex E (30m)].In the hydrogenation of MEA, [Ir(COD)Cl]<sub>2</sub> is used as metal complex and catalyst precursor. The hydrogenation is carried out in bulk using 105 g of MEA (without solvent) in the presence of 70 mg of tetrabutylammonium iodide and 10 ml of acetic acid.</p> <p><b>With</b> hydrogen, di(norbornadiene)rhodium(I) tetrafluoroborate, C<sub>62</sub>H<sub>78</sub>FeP<sub>4</sub> in methanol, Time= 1h, T= 25 °C , p= 750.075Torr , Product distribution / selectivity</p> <p><b>Patent; SOLVIAS AG;</b> WO2006/3195; (2006); (A1) English  <a href="#">View in Reaxys</a></p>
	<p>Example Name B.1  The method of carrying out the hydrogenations and the determination of the optical yields ee is described in general terms by W. Weissensteiner et al in Organometallics 21 (2002), pages 1766-1774. The catalysts are in each case prepared in situ in the solvent by mixing of the ligand and metal complex as catalyst precursor (unless indicated otherwise = [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub>). Unless indicated otherwise, the substrate concentration is 0.25 mol/l. The determination of conversion and ee of MAA is carried out by means of gas chromatography using a chiral column (Chirasil-L-val).The hydrogenations of EAC are carried out in ethanol in the presence of 5percent (v/v) of CF<sub>3</sub>CH<sub>2</sub>OH. The determination of the ee is carried out by means of gas chromatography using a chiral column [Lipodex E (30m); 1300C isothermal; 190 KPa H<sub>2</sub>]. In the hydrogenation of EOv, [Ru<sub>2</sub>(p-cumene)]<sub>2</sub> is used as metal complex and catalyst precursor. The determination of the ee is carried out after reaction with trifluoroacetic anhydride by means of gas chromatography using a chiral column [Lipodex E (30m)].In the hydrogenation of MEA, [Ir(COD)Cl]<sub>2</sub> is used as metal</p>

complex and catalyst precursor. The hydrogenation is carried out in bulk using 105 g of MEA (without solvent) in the presence of 70 mg of tetrabutylammonium iodide and 10 ml of acetic acid.

**With** hydrogen, triethylamine, di(norbornadiene)rhodium(I) tetrafluoroborate, C<sub>64</sub>H<sub>60</sub>FeN<sub>2</sub>P<sub>4</sub> in tetrahydrofuran, Time= 1h, T= 25 °C , p= 3750.38Torr , Product distribution / selectivity

**Patent:** SOLVIAS AG; WO2006/3195; (2006); (A1) English

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Yield	Conditions & References
86.50 %	<p><b>With</b> magnesium, methanol</p> <p><b>Verma, Raman K.; Singla, Rubina; Punniyakoti, V. T.;</b> Medicinal Chemistry Research; <b>vol.</b> 13; nb. 8-9; (2004); p. 660 - 676</p> <p><a href="#">View in Reaxys</a></p>
68.2 %	<p><b>With</b> H<sub>2</sub>, KOH, RhCl(PPh<sub>3</sub>)<sub>2</sub>-(Ph<sub>2</sub>PO<sub>2</sub>CCH=CMe<sub>2</sub>) in acetone, Time= 17h, T= 22 °C , p= 2280Torr</p> <p><b>Preston, Sheila A.; Cupertino, Domenico C.; Palma-Ramirez, Pilar; Cole-Hamilton, David J.;</b> Journal of the Chemical Society, Chemical Communications; nb. 12; (1986); p. 977 - 978</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, NEt<sub>3</sub>, &lt;Rh(PPh<sub>3</sub>)<sub>2</sub>(Ph<sub>2</sub>PO<sub>2</sub>CCMe=CHPh)&gt;PF<sub>6</sub> in tetrahydrofuran, Time= 20h, p= 2280Torr , Ambient temperature, various catalysts, other bases; other substituted propenoic acids and hexene, Product distribution, Mechanism</p> <p><b>Iraqi, Ahmed; Fairfax, Neil R.; Preston, Sheila A.; Cupertino, Domenico C.; Irvine, Derek J.; Cole-Hamilton, David J.;</b> Journal of the Chemical Society, Dalton Transactions: Inorganic Chemistry (1972-1999); nb. 8; (1991); p. 1929 - 1936</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> ammonium formate, Rh(PPh<sub>3</sub>)<sub>3</sub>Cl in dimethylsulfoxide, H<sub>2</sub>O, Time= 0.0833333h, Irradiation, also without microwave irradiation in var. temp.; also 2H labelled formate salts; also cinnamic acid, Product distribution</p> <p><b>Al-Qahtani, Mohammed H.; Cleator, Nicola; Danks, Timothy N.; Garman, Russell N.; Jones, John R.; et al.;</b> Journal of Chemical Research, Synopses; nb. 7; (1998); p. 400 - 401</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> diluted alcohol, sodium amalgam</p> <p><b>Conrad; Bischoff;</b> Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 180</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> aq. NaOH, Raney-Ni</p> <p><b>Watson,M.B.; Youngson,G.W.;</b> Journal of the Chemical Society [Section] C: Organic; (1968); p. 258 - 262</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>, NEt<sub>3</sub>, &lt;RhCl(PPh<sub>3</sub>)<sub>2</sub>(Ph<sub>2</sub>PO<sub>2</sub>CCMe=CHPh)&gt; in tetrahydrofuran, Time= 20h, p= 2280Torr , Ambient temperature, Yield given</p> <p><b>Iraqi, Ahmed; Fairfax, Neil R.; Preston, Sheila A.; Cupertino, Domenico C.; Irvine, Derek J.; Cole-Hamilton, David J.;</b> Journal of the Chemical Society, Dalton Transactions: Inorganic Chemistry (1972-1999); nb. 8; (1991); p. 1929 - 1936</p> <p><a href="#">View in Reaxys</a></p>
	<p>Example Name 60.b</p> <p>Example Title b</p> <p>b</p>

### $\alpha$ -Methyl-hydrocinnamic acid

A solution of  $\alpha$ -methyl cinnamic acid (10.0 g., 61.7 mmol.) in dry methanol (250 ml.) was treated with 10percent palladium on carbon and hydrogenated (balloon used) at room temperature for 16 hours.

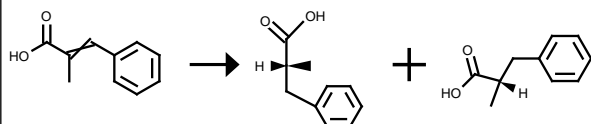
The reaction mixture was diluted with methanol (250 ml.), filtered through a Celite pad in a millipore unit, washing the pad well with methanol (2\*100 ml.).

The clear filtrate was evaporated to dryness to give 10.225 g. of title product as a thick syrup.

**With palladium in methanol**

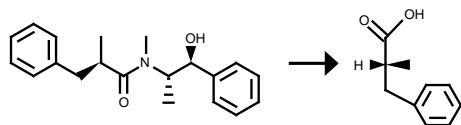
**Patent; E. R. Squibb and Sons, Inc.;** US5552397; (1996); (A1) English

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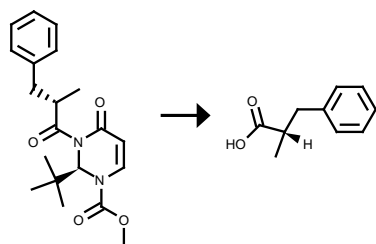


Rx-ID: 28260217 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>Example Name D.3</b> 0.41 g (2.53 mmol) of 2-methylcinnamic acid and 5 ml of degassed methanol are introduced in succession into a Schlenk vessel filled with argon. A catalyst solution consisting of 4.73 mg (0.01265 mmol) of [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub>, 8.77 mg (0.0133 mmol) of ligand A and 5 ml of degassed methanol is prepared in a second Schlenk vessel filled with argon. This solution and the catalyst solution are then transferred in succession by means of a steel capillary into a 50 ml autoclave filled with argon. The ratio of substrate/catalyst is 200. The autoclave is closed and a pressure of 5 bar is set by means of 4 flushing cycles (pressurization to 10 bar of hydrogen). The autoclave is thermostated at 25.deg. C. and the reaction is started by switching on the stirrer. The reactor is stirred for 19 hours. After opening the autoclave, a reddish reaction solution is isolated. The conversion is quantitative (determined by means of GC and <sup>1</sup>H-NMR). Removal of the solvent on a rotary evaporator gives a quantitative yield of 2-methyl-3-phenylpropionic acid having an enantiomeric purity of 29percent ee (determined by means of HPLC after conversion into the methyl ester; column: Chiracel OB.).</p> <p><b>With hydrogen, (S)-Fc-<math>\alpha</math>-S(C<sub>5</sub>H<sub>5</sub>)Fe(C<sub>5</sub>H<sub>3</sub>(PPh<sub>2</sub>)CH(OH)C<sub>6</sub>H<sub>4</sub>(PPh<sub>2</sub>)), [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub> in methanol, Time= 19h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</b></p> <p><b>Patent; Lotz, Matthias; Spindler, Felix;</b> US2008/287698; (2008); (A1) English <a href="#">View in Reaxys</a></p>
	<p><b>Example Name D.3</b> A method similar to that of Example D3 is employed. 8.96 mg (0.0133 mmol) of methoxy-, TANIAPHOS are used in place of the ligand A. The conversion is 100percent. Removal of the solvent on a rotary evaporator gives a quantitative yield of 2-methyl-3-phenylpropionic acid having an enantiomeric purity of 30percent ee.</p> <p><b>With hydrogen, (S)-1-diphenylphosphino-2-[<math>\alpha</math>-(S)-methoxy(o-diphenylphosphinophenyl)methyl]ferrocene, [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub> in methanol, Time= 19h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</b></p> <p><b>Patent; Lotz, Matthias; Spindler, Felix;</b> US2008/287698; (2008); (A1) English <a href="#">View in Reaxys</a></p>
	<p><b>With diphenyl{1-[(S)-2,6-bis-trimethylsilyl-3,5-dioxa-4-phospha-cyclohepta(2,1-a;3,4-a')dinaphthalen-4-yl]-3-methyl-2-indolyl}phosphine, di(norbornadiene)rhodium(I) tetrafluoroborate, hydrogen in dichloromethane, Time= 20h, T= 20 °C , p= 7500.75Torr , Inert atmosphere, optical yield given as percent ee</b></p> <p><b>Wassenaar, Jeroen; Reek, Joost N. H.; Kuil, Mark;</b> Advanced Synthesis and Catalysis; <b>vol.</b> 350; nb. 10; (2008); p. 1610 - 1614 <a href="#">View in Reaxys</a></p>


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Yield	Conditions & References
95 %	<p><b>With 9 N aq. H<sub>2</sub>SO<sub>4</sub> in dioxane, Time= 6h, Heating</b></p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinstry, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>
87 %	<p><b>With 18 N H<sub>2</sub>SO<sub>4</sub> in dioxane, Heating</b></p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 116; nb. 20; (1994); p. 9361 - 9362  <a href="#">View in Reaxys</a></p>
83 %	<p><b>With H<sub>2</sub>SO<sub>4</sub> in dioxane, Time= 1h, Heating</b></p> <p><b>Bach, Jordi; Galobardes, Marta; Garcia, Jordi; Romea, Pedro; Tey, Cristina; et al.;</b> Tetrahedron Letters; <b>vol.</b> 39; nb. 37; (1998); p. 6765 - 6768  <a href="#">View in Reaxys</a></p>

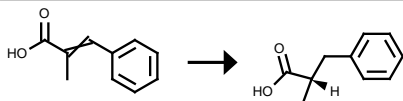

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Yield	Conditions & References
77 %	<p><b>With 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH in tetrahydrofuran, Time= 0.333333h</b></p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: aq. NaOH / methanol / 0.67 h / 4 °C            2: 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 2.) methyl chloroformate / 1.) THF, 30 min, 2.) THF, RT, 4 h  <b>With aq. NaOH, 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 2.) methyl chloroformate in methanol</b></p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: aq. NaOH / methanol / 0.67 h / 4 °C            2: 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH / 1.) THF, 30 min, 2.) THF, RT, 4 h  <b>With aq. NaOH, 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH in methanol</b></p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2</p>



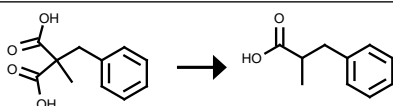
1: 80percent methanolic KOH / 0.33 h / 0 °C  
 2: LiOOH  
**With** 80percent methanolic KOH, LiOOH

**Negrete, George R.; Konopelski, Joseph P.**; Tetrahedron: Asymmetry; **vol.** 2; nb. 2; (1991); p. 105 - 108  
[View in Reaxys](#)



Rx-ID: 22694575 [View in Reaxys](#)

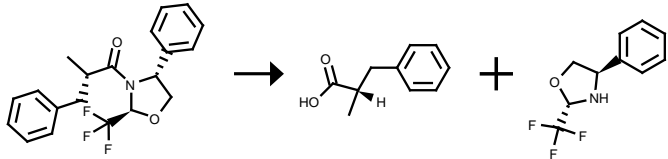
Yield	Conditions & References
99 %	<p><b>With</b> hydrogen, triethylamine in methanol, Time= 0.5h, T= 20 °C , p= 4560.31Torr , optical yield given as percent ee, enantioselective reaction</p> <p><b>Li, Shen; Zhu, Shou-Fei; Zhang, Can-Ming; Song, Song; Zhou, Qi-Lin</b>; Journal of the American Chemical Society; <b>vol.</b> 130; nb. 27; (2008); p. 8584 - 8585  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: (i) H<sub>2</sub>SO<sub>4</sub>, (ii) H<sub>2</sub>, Pd-C, EtOH            2: α-chymotrypsin, aq. NaOH  <b>With</b> α-chymotrypsin, aq. NaOH</p> <p><b>Cohen, S.G.; Milovanovic, A.</b>; Journal of the American Chemical Society; <b>vol.</b> 90; (1968); p. 3495 - 3502  <a href="#">View in Reaxys</a></p>

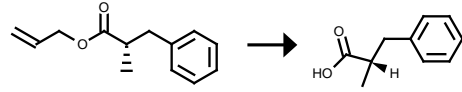


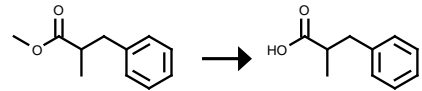
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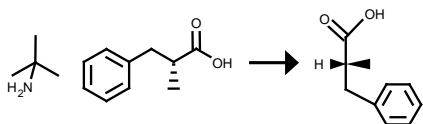
Yield	Conditions & References
95 %	<p><b>With</b> copper(I) oxide in acetonitrile, Time= 6h, Heating</p> <p><b>Toussaint, Olivier; Capdevielle, Patrice; Maumy, Michel</b>; Synthesis; nb. 12; (1986); p. 1029 - 1031  <a href="#">View in Reaxys</a></p>
	<p><b>Conrad; Bischoff</b>; Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 178  <a href="#">View in Reaxys</a></p>
	<p>(decarboxylation)</p> <p><b>Terashima, S. et al.</b>; Chemical and Pharmaceutical Bulletin; <b>vol.</b> 18; (1970); p. 1124 - 1136  <a href="#">View in Reaxys</a></p>
	<p>in dimethylformamide, Time= 4h, T= 100 - 110 °C</p> <p><b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.</b>; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> conc. H<sub>2</sub>SO<sub>4</sub>, Time= 1h, Heating, Yield given</p> <p><b>Tyrrell, Elizabeth; Tsang, Michael W. H.; Skinner, George A.; Fawcett, John</b>; Tetrahedron; <b>vol.</b> 52; nb. 29; (1996); p. 9841 - 9852  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> H<sub>2</sub>SO<sub>4</sub>, Time= 3h, Heating</p> <p><b>Dib, H. H.; Ibrahim, M. R.; Al-Awadi, N. A.; Ibrahim, Y. A.; Al-Awadi, S.</b>; International Journal of Chemical Kinetics; <b>vol.</b> 40; nb. 2; (2008); p. 51 - 58</p>

	<a href="#">View in Reaxys</a>
16.5 g	Time= 2h, T= 170 °C  <b>Alcalde, Ermitas; Mesquida, Neus; Lopez-Perez, Sara; Frigola, Jordi; Merce, Ramon</b> ; Organic and Biomolecular Chemistry; <b>vol.</b> 6; nb. 20; (2008); p. 3795 - 3810 <a href="#">View in Reaxys</a>

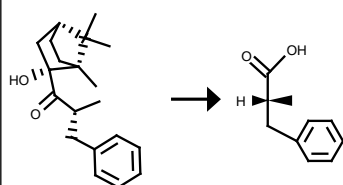
	
Rx-ID: 11214120 <a href="#">View in Reaxys</a>	
Yield	Conditions & References
97 %, 93 %	<b>Stage 1: With LiAlH<sub>4</sub> in diethyl ether, Time= 0.5h, T= -10 °C</b> <b>Stage 2: With aq. NaCl in diethyl ether, T= -10 - 20 °C</b> <b>Stage 3: With 2-methyl-2-butene, sodium dihydrogen phosphate monohydrate, sodium chlorite in tetrahydrofuran, H<sub>2</sub>O, 2-methyl-propan-2-ol, Time= 1.2h, T= 20 °C , Further stages.</b>  <b>Tessier, Arnaud; Pytkowicz, Julien; Brigaud, Thierry</b> ; Angewandte Chemie, International Edition; <b>vol.</b> 45; nb. 22; (2006); p. 3677 - 3681; Angewandte Chemie; <b>vol.</b> 118; nb. 22; (2006); p. 3759 - 3763 <a href="#">View in Reaxys</a>

	
Rx-ID: 3231137 <a href="#">View in Reaxys</a>	
Yield	Conditions & References
89 %	<b>With H<sub>2</sub>O, &lt;RhCl(PPh<sub>3</sub>)<sub>3</sub>&gt; in ethanol, Time= 5h, Heating</b>  <b>Oppolzer, Wolfgang; Lienard, Philippe</b> ; Helvetica Chimica Acta; <b>vol.</b> 75; nb. 8; (1992); p. 2572 - 2582 <a href="#">View in Reaxys</a>

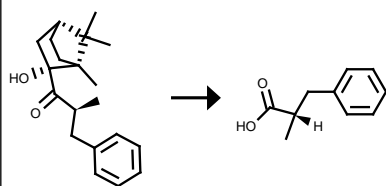
	
Rx-ID: 2042017 <a href="#">View in Reaxys</a>	
Yield	Conditions & References
100 %	<b>With potassium hydroxide in ethanol, H<sub>2</sub>O, Time= 8h, T= 0 °C</b>  <b>Kato, Dai-ichiro; Mitsuda, Satoshi; Ohta, Hiromichi</b> ; Journal of Organic Chemistry; <b>vol.</b> 68; nb. 19; (2003); p. 7234 - 7242 <a href="#">View in Reaxys</a>
95 %	<b>With aq. NaOH in methanol, Time= 1h, Heating</b>  <b>Smonou, Ioulia; Orfanopoulos, Michael</b> ; Synthetic Communications; <b>vol.</b> 20; nb. 9; (1990); p. 1387 - 1397 <a href="#">View in Reaxys</a>


Rx-ID: 10652662 [View in Reaxys](#)

Yield	Conditions & References
	<p>Example Name 12.i</p> <p>(λ)-2-Methy-3-phenylpropionic acid ter/-butylamine salt, reaction time 40 minutes, conversion 100percent. The free acid was liberated by partitioning between dichloromethane and 2M HCl. The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent was evaporated, then the product was distilled (kugelrohr, 150 °C, 0.5 mbar) to give (λ)-2-Methy-3- phenylpropionic acid as a colorless liquid, [α]<sub>D</sub><sup>20</sup> -22.1, c = 1.02, CHCl<sub>3</sub>. Lit. (E. Tyrell, M.W.H. Tang, G.A. Skinner and J. Fawcett, Tetrahedron, 1996, 52, 9841-9852, [α]<sub>D</sub><sup>20</sup> -23.1 c = 1, CHCl<sub>3</sub>). ee 85percent (derivatised using TMS-diazomethane, Chirasil Dex CB column, 25m x 0.25 mm, injector/detector 200 °C, helium 20 psi, 100 °C for 21 minutes then ramp at 15 .deg.C/min to 200 °C, hold for 5 minutes, retention times R 30.40 minutes, S 31.06 minutes, (E)-methyl 2-methylcinnamate, 34.79 minutes). <sup>1</sup>H NMR analysis of the (S)- &lt;n="32"/&gt;methyl mandelate (E. Tyrell, M.W.H. Tang, G.A. Skinner and J. Fawcett, Tetrahedron, 1996, 52, 9841 -9852) confirmed assignment of (λ)-configuration.</p> <p><b>With hydrogen chloride in dichloromethane, water</b></p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.;</b> WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 8912508 [View in Reaxys](#)

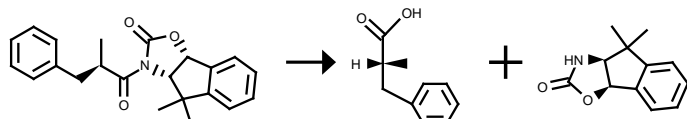
Yield	Conditions & References
95 %	<p><b>With CAN in acetonitrile, H<sub>2</sub>O, Time= 1h, T= 0 °C</b></p> <p><b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Lecumberri, Ainara; Linden, Anthony;</b> Organic Letters; <b>vol. 3;</b> nb. 21; (2001); p. 3249 - 3252</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 8912509 [View in Reaxys](#)

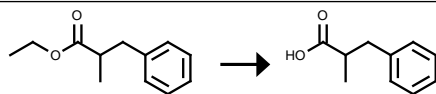
Yield	Conditions & References
90 %	<p><b>With CAN in acetonitrile, H<sub>2</sub>O, Time= 1h, T= 0 °C</b></p> <p><b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Lecumberri, Ainara; Linden, Anthony;</b> Organic Letters; <b>vol. 3;</b> nb. 21; (2001); p. 3249 - 3252</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2147536 [View in Reaxys](#)

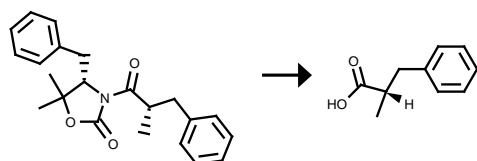
Yield	Conditions & References
90 %	<p><b>With</b> chromic acid, water in diethyl ether, Time= 2h, T &lt;25 deg C</p> <p><b>Rangaishenvi, Milind V.; Singaram, Bakthan; Brown, Herbert C.;</b> Journal of Organic Chemistry; <b>vol.</b> 56; nb. 10; (1991); p. 3286 - 3294</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 4368729 [View in Reaxys](#)

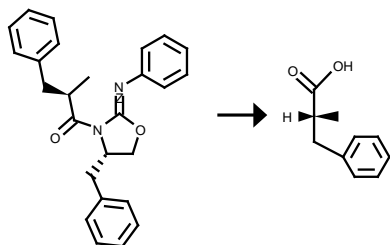
Yield	Conditions & References
100 %, 95 %	<p><b>With</b> aq. LiOOH in tetrahydrofuran, T= 0 °C</p> <p><b>Sudo, Atsushi; Saigo, Kazuhiko;</b> Tetrahedron: Asymmetry; <b>vol.</b> 6; nb. 9; (1995); p. 2153 - 2156</p> <p><a href="#">View in Reaxys</a></p>
100 %, 95 %	<p><b>With</b> aq. H<sub>2</sub>O<sub>2</sub>, LiOH in tetrahydrofuran, Time= 2.5h, T= 0 °C</p> <p><b>Sudo, Atsushi; Saigo, Kazuhiko;</b> Tetrahedron: Asymmetry; <b>vol.</b> 7; nb. 10; (1996); p. 2939 - 2956</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 1992712 [View in Reaxys](#)

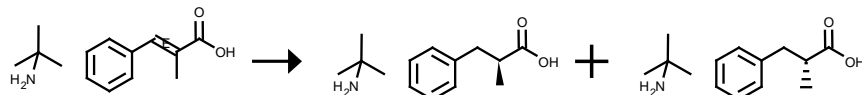
Yield	Conditions & References
74 %	<p><b>With</b> H<sub>2</sub>O/OH- in ethanol, Time= 5h, Heating</p> <p><b>Knorr, Rudolf; Lattke, Ernst;</b> Chemische Berichte; <b>vol.</b> 114; nb. 6; (1981); p. 2116 - 2131</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> aq. KOH</p> <p><b>Crombie, Aimee L.; Kane, John L.; Shea, Kevin M.; Danheiser, Rick L.;</b> Journal of Organic Chemistry; <b>vol.</b> 69; nb. 25; (2004); p. 8652 - 8667</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> KOH, H<sub>2</sub>O</p> <p><b>Kane, John L.; Shea, Kevin M.; Crombie, Aimee L.; Danheiser, Rick L.;</b> Organic Letters; <b>vol.</b> 3; nb. 7; (2001); p. 1081 - 1084</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 4201083 [View in Reaxys](#)

Yield	Conditions & References
100 %	<p><b>With</b> LiOH, water <b>in</b> tetrahydrofuran, T= 0 - 25 °C</p> <p><b>Davies, Stephen G.; Sangane, Hitesh J.</b>; Tetrahedron: Asymmetry; <b>vol.</b> 6; nb. 3; (1995); p. 671 - 674</p> <p><a href="#">View in Reaxys</a></p>

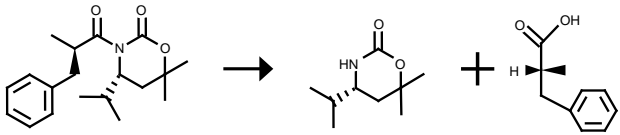

Rx-ID: 9043914 [View in Reaxys](#)

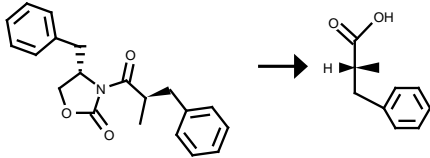
Yield	Conditions & References
96 %	<p><b>With</b> aq. NaOH <b>in</b> dioxane, Time= 0.5h, Heating</p> <p><b>Lee, Gue-Jae; Kim, Taek Hyeon; Kim, Jae Nyoung; Lee, Uk</b>; Tetrahedron: Asymmetry; <b>vol.</b> 13; nb. 1; (2002); p. 9 - 12</p> <p><a href="#">View in Reaxys</a></p>

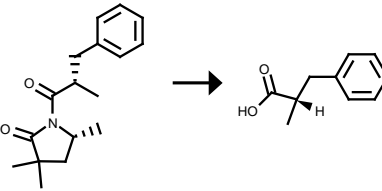

Rx-ID: 10652632 [View in Reaxys](#)

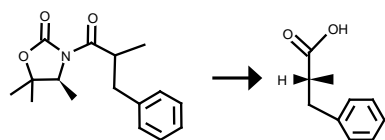
Yield	Conditions & References
	<p>Example Name 12.v</p> <p><b>With</b> hydrogen, [((R,R)-Me-5-Fc)Rh(COD)]<sup>(1+)</sup> <b>in</b> methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.</b>; WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>
	<p>Example Name 12.v</p> <p><b>With</b> hydrogen, [((R,R)-Et-5-Fc)Rh(COD)]<sup>(1+)</sup> <b>in</b> methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.</b>; WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>
	<p>Example Name 12.v</p> <p><b>With</b> hydrogen, [((R,R)-Ph-BPE)Rh(COD)]<sup>(1+)</sup> <b>in</b> methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.</b>; WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>
	<p>Example Name 12.v</p> <p><b>With</b> hydrogen, [((R,R)-iPr-5-Fc)Rh(COD)]<sup>(1+)</sup> <b>in</b> methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.</b>; WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>

	<p><b>With</b> hydrogen, [(bis(diphenylphospholanyl)ferrocene)Rh(cyclooctadiene)]BF<sub>4</sub> <b>in</b> methanol, Time= 1h, T= 25 °C , p= 7500.75Torr , Title compound not separated from byproducts.</p> <p><b>Fox, Martin E.; Jackson, Mark; Lennon, Ian C.; Klosin, Jerzy; Abboud, Khalil A.;</b> Journal of Organic Chemistry; <b>vol.</b> 73; nb. 3; (2008); p. 775 - 784  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> hydrogen, [(bis(di-iPr-phospholanyl)ferrocene)Rh(cyclooctadiene)]BF<sub>4</sub> <b>in</b> methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Title compound not separated from byproducts.</p> <p><b>Fox, Martin E.; Jackson, Mark; Lennon, Ian C.; Klosin, Jerzy; Abboud, Khalil A.;</b> Journal of Organic Chemistry; <b>vol.</b> 73; nb. 3; (2008); p. 775 - 784  <a href="#">View in Reaxys</a></p>

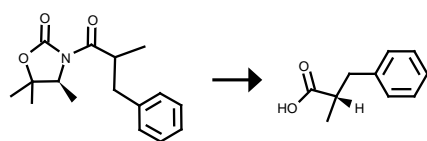
		Rx-ID: 11137427 <a href="#">View in Reaxys</a>
Yield	Conditions & References	
100 %, 12 mg	<p><b>With</b> aq. H<sub>2</sub>O<sub>2</sub>, LiOH <b>in</b> tetrahydrofuran, Time= 16h, T= 0 - 20 °C</p> <p><b>Davies, Stephen G.; Garner, A. Christopher; Roberts, Paul M.; Smith, Andrew D.; Sweet, Miles J.; Thomson, James E.;</b> Organic and Biomolecular Chemistry; <b>vol.</b> 4; nb. 14; (2006); p. 2753 - 2768  <a href="#">View in Reaxys</a></p>	

		Rx-ID: 4493212 <a href="#">View in Reaxys</a>
Yield	Conditions & References	
78 %	<p><b>With</b> aq. LiOH, H<sub>2</sub>O<sub>2</sub> <b>in</b> tetrahydrofuran, Time= 1h, T= 0 °C</p> <p><b>Tyrrell, Elizabeth; Tsang, Michael W. H.; Skinner, George A.; Fawcett, John;</b> Tetrahedron; <b>vol.</b> 52; nb. 29; (1996); p. 9841 - 9852  <a href="#">View in Reaxys</a></p>	

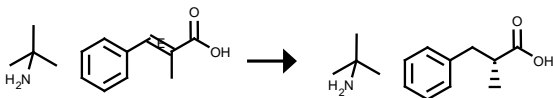
		Rx-ID: 9026989 <a href="#">View in Reaxys</a>
Yield	Conditions & References	
83 %	<p><b>With</b> aq. LiOH <b>in</b> tetrahydrofuran, T= 0 °C</p> <p><b>Davies, Stephen G.; Dixon, Darren J.; Doisneau, Gilles J.-M.; Prodger, Jeremy C.; Sangane, Hitesh J.;</b> Tetrahedron: Asymmetry; <b>vol.</b> 13; nb. 6; (2002); p. 647 - 658  <a href="#">View in Reaxys</a></p>	


Rx-ID: 10462650 [View in Reaxys](#)

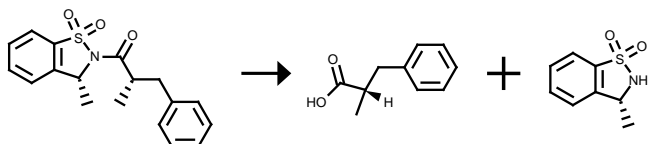
Yield	Conditions & References
77 %	<p><b>With</b> aq. LiOH in tetrahydrofuran, T= 0 - 20 °C</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki;</b> Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 10462651 [View in Reaxys](#)

Yield	Conditions & References
78 %	<p><b>With</b> aq. LiOH in tetrahydrofuran, T= 0 - 20 °C</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki;</b> Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 10652623 [View in Reaxys](#)

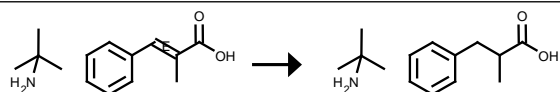
Yield	Conditions & References
	<p>Example Name 12.i; 12.iv</p> <p>(Λ)-2-Methy-3-phenylpropionic acid ter-/butylamine salt, reaction time 40 minutes, conversion 100percent. The free acid was liberated by partitioning between dichloromethane and 2M HCl. The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent was evaporated, then the product was distilled (kugelrohr, 150 °C, 0.5 mbar) to give (Λ)-2-Methy-3-phenylpropionic acid as a colorless liquid, [α]<sub>D</sub><sup>20</sup> -22.1, c = 1.02, CHCl<sub>3</sub>. Lit. (E. Tyrell, M.W.H. Tang, G.A. Skinner and J. Fawcett, Tetrahedron, 1996, 52, 9841-9852, [α]<sub>D</sub><sup>20</sup> -23.1 c = 1, CHCl<sub>3</sub>). ee 85percent (derivatised using TMS-diazomethane, Chirasil Dex CB column, 25m x 0.25 mm, injector/detector 200 °C, helium 20 psi, 100 °C for 21 minutes then ramp at 15 .deg.C/min to 200 °C, hold for 5 minutes, retention times R 30.40 minutes, S 31.06 minutes, (E)-methyl 2-methylcinnamate, 34.79 minutes). <sup>1</sup>H NMR analysis of the (S)- &lt;n="32"/&gt;methyl mandelate (E. Tyrell, M.W.H. Tang, G.A. Skinner and J. Fawcett, Tetrahedron, 1996, 52, 9841 -9852) confirmed assignment of (Λ-configuration.</p> <p><b>With</b> hydrogen, 1,1'-bis[(2S,5S)-diphenylphospholano]ferrocene(1,5-cyclooctadiene)rhodium(I) tetrafluoroborate in methanol, Time= 0.666667h, T= 25 - 30 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.;</b> WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>
	<p>Example Name 12.iv</p> <p><b>With</b> hydrogen, 1,1'-bis[(2S,5S)-diphenylphospholano]ferrocene(1,5-cyclooctadiene)rhodium(I) tetrafluoroborate in methanol, Time= 0.666667h, T= 30 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.;</b> WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2474734 [View in Reaxys](#)

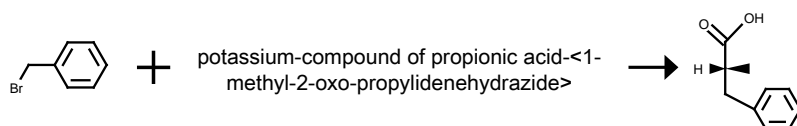
Yield	Conditions & References
81 %	<p><b>With</b> aq. H<sub>2</sub>O<sub>2</sub> (30 percent), LiOH·H<sub>2</sub>O in tetrahydrofuran, H<sub>2</sub>O, T= 0 °C</p> <p><b>Oppolzer, Wolfgang; Rodriguez, Ines; Starkemann, Christian; Walther, Eric;</b> Tetrahedron Letters; <b>vol.</b> 31; nb. 35; (1990); p. 5019 - 5022</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 4818648 [View in Reaxys](#)

Yield	Conditions & References
89 %	<p><b>With</b> H<sub>3</sub>O<sup>+</sup> in tetrahydrofuran, T= 100 °C</p> <p><b>Suprun, W. Y.;</b> Journal fuer Praktische Chemie/Chemiker-Zeitung; <b>vol.</b> 340; nb. 3; (1998); p. 247 - 255</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 10652636 [View in Reaxys](#)

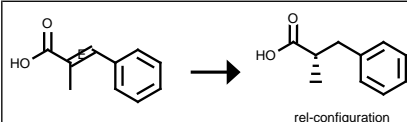
Yield	Conditions & References
	<p>Example Name 12.v</p> <p><b>With</b> hydrogen, (S,S)-2,3-bis(2,5-diphenyl-phospholan-1-yl)-(1,5-cyclooctadiene)rhodium(I) tetrafluoroborate in methanol, Time= 18h, T= 25 °C , p= 7500.75Torr , Product distribution / selectivity</p> <p><b>Patent; DOW GLOBAL TECHNOLOGIES INC.;</b> WO2007/123957; (2007); (A2) English</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> hydrogen, [bis(diphenylphospholanyl)quinoxaline-Rh(cyclooctadiene)]BF<sub>4</sub> in methanol, Time= 18h, T= 25 °C , p= 7500.75Torr</p> <p><b>Fox, Martin E.; Jackson, Mark; Lennon, Ian C.; Klosin, Jerzy; Abboud, Khalil A.;</b> Journal of Organic Chemistry; <b>vol.</b> 73; nb. 3; (2008); p. 775 - 784</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 16745184 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 82 percent / LDA, LiCl / tetrahydrofuran / 0.75 h / 0 °C</p> <p>2: 83 percent / H<sub>2</sub>SO<sub>4</sub> / dioxane / 1 h / Heating</p> <p><b>With</b> LDA, LiCl, H<sub>2</sub>SO<sub>4</sub> in tetrahydrofuran, dioxane</p>

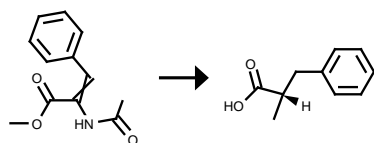


	<p><b>Bach, Jordi; Galobardes, Marta; Garcia, Jordi; Romea, Pedro; Tey, Cristina; et al.;</b> Tetrahedron Letters; <b>vol.</b> 39; nb. 37; (1998); p. 6765 - 6768</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: 90 percent / NaHMDS            2: LiOH, H<sub>2</sub>O<sub>2</sub>  <b>With</b> NaHMDS, LiOH, H<sub>2</sub>O<sub>2</sub></p> <p><b>Simpson, Thomas J.; Smith, Robert W.; Westaway, Susan M.; Willis, Christine L.; Buss, Antony D.; et al.;</b> Tetrahedron Letters; <b>vol.</b> 38; nb. 30; (1997); p. 5367 - 5370</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: 90 percent / LiCl, diisopropylamine, n-BuLi / tetrahydrofuran; hexane / 0.25 h / 0 °C            2: 95 percent / 9 N aq. H<sub>2</sub>SO<sub>4</sub> / dioxane / 6 h / Heating  <b>With</b> LiCl, diisopropylamine, n-BuLi, 9 N aq. H<sub>2</sub>SO<sub>4</sub> in tetrahydrofuran, hexane, dioxane</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinsty, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511</p> <p><a href="#">View in Reaxys</a></p>



Rx-ID: 25598370 [View in Reaxys](#)

Yield	Conditions & References
	<p>Example Name D.10</p> <p>The experimental procedure is similar to that of Example D1. 2.53 mmol of starting material are always used, and the ratio of substrate to catalyst (s/c) is always 200. The reaction parameters and the results are summarized in Table 1 below. In Examples 5, 11, 16 and 20, toluene (10 ml) is used as solvent. In Examples 6 to 8, 12, 13, 17, 18, 21 and 22, ethanol (10 ml, in Examples 6 and 17 9.5 ml) is used as solvent. In Examples 9, 10, 14, 15 and 19, methanol (10 ml) is used as solvent.</p> <p><b>With</b> hydrogen, di(norbornadiene)rhodium(I) tetrafluoroborate, (S)-1-di(3,5-dimethyl-4-methoxyphenyl)phosphino-2-[α-(S)-hydroxy(o-diphenylphosphinophenyl)methyl]ferrocene in methanol, Time= 19h, T= 25 °C, p= 3750.38Torr, Product distribution / selectivity</p> <p><b>Patent:</b> UMICORE AG and CO. KG; WO2005/108409; (2005); (A2) English</p> <p><a href="#">View in Reaxys</a></p>

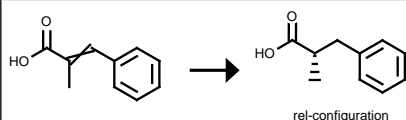


Rx-ID: 28260214 [View in Reaxys](#)

Yield	Conditions & References
	<p>Example Name D.14</p> <p>The experimental procedure is similar to that of Example D1. 2.53 mmol of starting material are always used, and the ratio of substrate to catalyst (sic) is always 200. The reaction parameters and the results are summarized in Table 1 below. In Examples 5, 11, 16 and 20, toluene (10 ml) is used as solvent. In Examples 6 to 8, 12, 13, 17, 18, 21 and 22, ethanol (10 ml, in Examples 6 and 17 9.5 ml) is used as solvent. In Examples 9, 10, 14, 15 and 19, methanol (10 ml) is used as solvent.</p> <p><b>With</b> hydrogen, (S)-1-di(3,5-dimethyl-4-methoxyphenyl)phosphino-2-[α-(S)-hydroxy(o-di(3,5-bis(trifluoromethyl)phenyl)phosphinophenyl)methyl]ferrocene, [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub> in methanol, Time= 1h, T= 25 °C, p= 750.075Torr, Product distribution / selectivity</p>

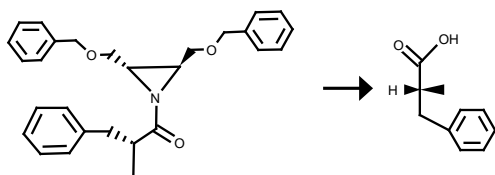
Patent; Lotz, Matthias; Spindler, Felix; US2008/287698; (2008); (A1) English

[View in Reaxys](#)



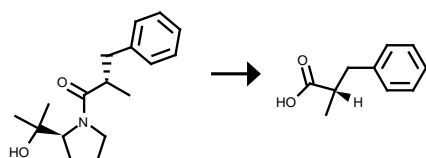
Rx-ID: 28260218 [View in Reaxys](#)

Yield	Conditions & References
	<p>Example Name D.10</p> <p>The experimental procedure is similar to that of Example DI. 2.53 mmol of starting material are always used, and the ratio of substrate to catalyst (sic) is always 200. The reaction parameters and the results are summarized in Table I below. In Examples 5, 11, 16 and 20, toluene (10 ml) is used as solvent. In Examples 6 to 8, 12, 13, 17, 18, 21 and 22, ethanol (10 ml, in Examples 6 and 17 9.5 ml) is used as solvent. In Examples 9, 10, 14, 15 and 19, methanol (10 ml) is used as solvent.</p> <p><b>With</b> hydrogen, (S)-1-di(3,5-dimethyl-4-methoxyphenyl)phosphino-2-[α-(S)-hydroxy(o-diphenylphosphinophenyl)methyl]ferrocene, [Rh(norbornadiene)<sub>2</sub>]BF<sub>4</sub> in methanol, Time= 19h, T= 25 °C , p= 3750.38Torr , Product distribution / selectivity</p> <p>Patent; Lotz, Matthias; Spindler, Felix; US2008/287698; (2008); (A1) English</p> <p><a href="#">View in Reaxys</a></p>



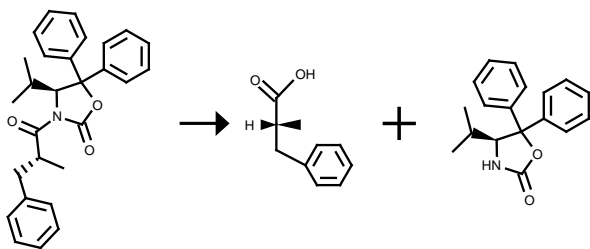
Rx-ID: 2515342 [View in Reaxys](#)

Yield	Conditions & References
63 %	<p><b>With</b> 30percent aq. H<sub>2</sub>O<sub>2</sub>, LiOH in tetrahydrofuran, H<sub>2</sub>O, Time= 120h, Ambient temperature</p> <p><b>Tanner, David; Birgersson, Carin; Gogoll, Adolf; Luthman, Kristina;</b> Tetrahedron; <b>vol.</b> 50; nb. 32; (1994); p. 9797 - 9824</p> <p><a href="#">View in Reaxys</a></p>

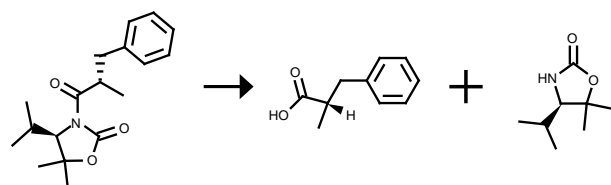


Rx-ID: 3499087 [View in Reaxys](#)

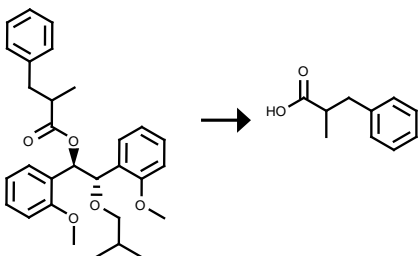
Yield	Conditions & References
76 %	<p><b>With</b> 3 M aq. HCl in dioxane, Time= 42h, T= 90 - 95 °C</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjorn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 5114767 [View in Reaxys](#)

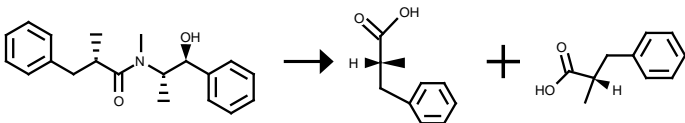
Yield	Conditions & References
79 %, 79 %	<p><b>With</b> LiOH·H<sub>2</sub>O, H<sub>2</sub>O<sub>2</sub> in H<sub>2</sub>O, tetrahydrofuran, Time= 1.5h, T= 0 °C</p> <p><b>Hintermann, Tobias; Seebach, Dieter</b>; Helvetica Chimica Acta; <b>vol.</b> 81; nb. 11; (1998); p. 2093 - 2126</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 5118676 [View in Reaxys](#)

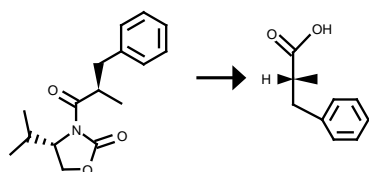
Yield	Conditions & References
96 %, 98 %	<p><b>With</b> LiOH·H<sub>2</sub>O, H<sub>2</sub>O in tetrahydrofuran, Time= 24h, Ambient temperature</p> <p><b>Bull, Steven D.; Davies, Stephen G.; Jones, Simon; Sanganee, Hitesh J.</b>; Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 4; (1999); p. 387 - 398</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 28576617 [View in Reaxys](#)

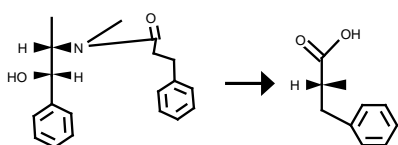
Yield	Conditions & References
82 %	<p><b>Stage 1:</b> With lithium hydroxide monohydrate, water in tetrahydrofuran, methanol, Time= 12h, T= 20 °C</p> <p><b>Stage 2:</b> With water, sodium hydrogencarbonate</p> <p><b>Stage 3:</b> With hydrogen chloride in water</p> <p><b>Broeker, Joachim; Knollmueller, Max; Gaertner, Peter</b>; Tetrahedron: Asymmetry; <b>vol.</b> 20; nb. 3; (2009); p. 273 - 287</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 4016116 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> tetra-n-butylammonium hydroxide in H<sub>2</sub>O, 2-methyl-propan-2-ol, Heating, Title compound not separated from byproducts</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 116; nb. 20; (1994); p. 9361 - 9362  <a href="#">View in Reaxys</a></p>
	<p><b>With</b> aq. n-Bu<sub>4</sub>NOH in 2-methyl-propan-2-ol, Heating, Yield given. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinstry, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>

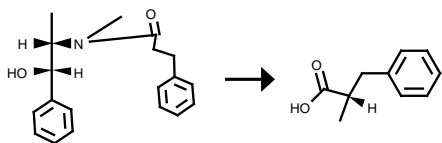

Rx-ID: 4833359 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> LiOH, H<sub>2</sub>O<sub>2</sub></p> <p><b>Simpson, Thomas J.; Smith, Robert W.; Westaway, Susan M.; Willis, Christine L.; Buss, Antony D.; et al.;</b> Tetrahedron Letters; <b>vol.</b> 38; nb. 30; (1997); p. 5367 - 5370  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: 93 percent / PhCH<sub>2</sub>OH / tetrahydrofuran / 1 h / 0 °C            2: hydrogen  <b>With</b> PhCH<sub>2</sub>OH, hydrogen in tetrahydrofuran</p> <p><b>Evans, D. A.; Ennis, M. D.; Mathre, D. J.;</b> Journal of the American Chemical Society; <b>vol.</b> 104; nb. 6; (1982); p. 1737 - 1739  <a href="#">View in Reaxys</a></p>


Rx-ID: 17169383 [View in Reaxys](#)

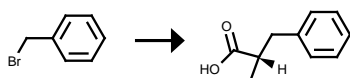
Yield	Conditions & References
	<p>Reaction Steps: 2            1: 95 percent / LiCl, diisopropylamine, n-BuLi / tetrahydrofuran; hexane / -78 °C            2: aq. n-Bu<sub>4</sub>NOH / 2-methyl-propan-2-ol / Heating  <b>With</b> LiCl, diisopropylamine, n-BuLi, aq. n-Bu<sub>4</sub>NOH in tetrahydrofuran, hexane, 2-methyl-propan-2-ol</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinstry, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: 1.) LDA, LiCl / 1.) THF, from -78 to 23 deg C, 80 min, 2.) THF, -78 deg C, 8 h            2: tetra-n-butylammonium hydroxide / H<sub>2</sub>O; 2-methyl-propan-2-ol / Heating  <b>With</b> 1.) LDA, LiCl, tetra-n-butylammonium hydroxide in H<sub>2</sub>O, 2-methyl-propan-2-ol</p>

**Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; Gleason, James L.;** Journal of the American Chemical Society; **vol.** 116; nb. 20; (1994); p. 9361 - 9362  
[View in Reaxys](#)



Rx-ID: 17169384 [View in Reaxys](#)

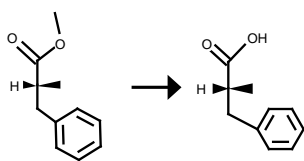
Yield	Conditions & References
	<p>Reaction Steps: 2            1: 95 percent / LiCl, diisopropylamine, n-BuLi / tetrahydrofuran; hexane / -78 °C            2: aq. n-Bu<sub>4</sub>NOH / 2-methyl-propan-2-ol / Heating  <b>With</b> LiCl, diisopropylamine, n-BuLi, aq. n-Bu<sub>4</sub>NOH in tetrahydrofuran, hexane, 2-methyl-propan-2-ol</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinsty, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2            1: 1.) LDA, LiCl / 1.) THF, from -78 to 23 deg C, 80 min, 2.) THF, -78 deg C, 8 h            2: tetra-n-butylammonium hydroxide / H<sub>2</sub>O; 2-methyl-propan-2-ol / Heating  <b>With</b> 1.) LDA, LiCl, tetra-n-butylammonium hydroxide in H<sub>2</sub>O, 2-methyl-propan-2-ol</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 116; nb. 20; (1994); p. 9361 - 9362  <a href="#">View in Reaxys</a></p>



Rx-ID: 20993453 [View in Reaxys](#)

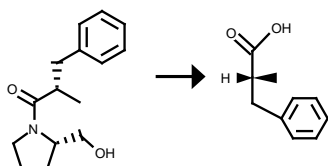
Yield	Conditions & References
	<p>Reaction Steps: 2            1: 1.) hexamethyldisilylamide / 1.) THF, -78 deg C, 2.5 h, 2.) THF, -23 deg C, 4 h            2: 77 percent / 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH / tetrahydrofuran / 0.33 h  <b>With</b> 1.) hexamethyldisilylamide, 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH in tetrahydrofuran</p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 3            1: 1.) hexamethyldisilylamide / 1.) THF, -78 deg C, 2.5 h, 2.) THF, -23 deg C, 4 h            2: aq. NaOH / methanol / 0.67 h / 4 °C            3: 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 2.) methyl chloroformate / 1.) THF, 30 min, 2.) THF, RT, 4 h  <b>With</b> 1.) hexamethyldisilylamide, aq. NaOH, 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 2.) methyl chloroformate in methanol</p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 3            1: 1.) hexamethyldisilylamide / 1.) THF, -78 deg C, 2.5 h, 2.) THF, -23 deg C, 4 h            2: aq. NaOH / methanol / 0.67 h / 4 °C            3: 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH / 1.) THF, 30 min, 2.) THF, RT, 4 h  <b>With</b> 1.) hexamethyldisilylamide, aq. NaOH, 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH in methanol</p>

**Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.**; Journal of the American Chemical Society; **vol.** 114; nb. 5; (1992); p. 1800 - 1812  
[View in Reaxys](#)



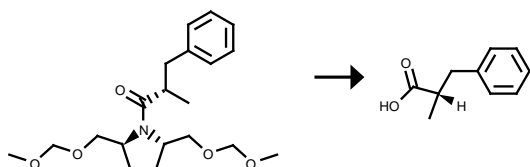
Rx-ID: 2199186 [View in Reaxys](#)

Yield	Conditions & References
92 %	<b>With</b> lipase from <i>Candida cylindracea</i> , Time= 4.5h, Ambient temperature, 0.1 M phosphate buffer, pH 7.0  <b>Delnick, Deborah L.; Margolin, Alexey L.</b> ; Tetrahedron Letters; <b>vol.</b> 31; nb. 47; (1990); p. 6797 - 6798 <a href="#">View in Reaxys</a>



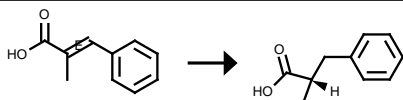
Rx-ID: 2658424 [View in Reaxys](#)

Yield	Conditions & References
92 %	<b>With</b> 1.0 N HCl, Time= 2h, T= 100 °C  <b>Evans, D. A.; Takacs, J. M.</b> ; Tetrahedron Letters; <b>vol.</b> 21; (1980); p. 4233 - 4236 <a href="#">View in Reaxys</a>



Rx-ID: 2700540 [View in Reaxys](#)

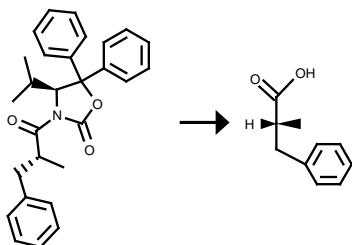
Yield	Conditions & References
90 %	<b>With</b> 1N HCl, Time= 3h, Heating  <b>Kawanami, Yasuhiro; Ito, Yoshio; Kitagawa, Toshiyuki; Taniguchi, Yoshiyuki; Katsuki, Tsutomu; Yamaguchi, Masaru</b> ; Tetrahedron Letters; <b>vol.</b> 25; nb. 8; (1984); p. 857 - 860 <a href="#">View in Reaxys</a>



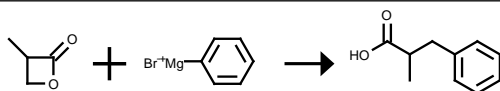
Rx-ID: 3962088 [View in Reaxys](#)

Yield	Conditions & References
51 % Spectr.	<b>With</b> H <sub>2</sub> , (S,R)-<(P-N)RhNBD>ClO <sub>4</sub> in ethanol, Time= 2.7h, T= 30 °C , study of the asymmetric hydrogenation of various prochiral substrates with chiral ligands containing rhodium(I) complexes of ferrocenylphosphonium as asymmetric catalysts, Product distribution  <b>Appelton, Trevor D.; Cullen, William R.; Evans, Stephen; Kim, Tae-Jeong; Trotter, James</b> ; Journal of Organometallic Chemistry; <b>vol.</b> 279; (1985); p. 5 - 22

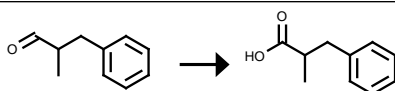
	<a href="#">View in Reaxys</a>
24 %Chromat.	<p><b>With</b> di(norbornadiene)rhodium(I) tetrafluoroborate, hydrogen <b>in</b> tetrahydrofuran, Time= 24h, T= 20 °C , p= 7500.75Torr , optical yield given as percent ee, enantioselective reaction</p> <p><b>Lee, Jong-Dae; Thanh, Thien Co; Kim, Tae-Jeong; Sang, Ook Kang;</b> Synlett; nb. 5; (2009); p. 771 - 774</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 8837301 [View in Reaxys](#)

Yield	Conditions & References
60 %	<p><b>With</b> aq. LiOH <b>in</b> tetrahydrofuran, Time= 25h, T= 0 - 24 °C</p> <p><b>Alexander, Karen; Cook, Stuart; Gibson, Colin L.; Kennedy, Alan R.;</b> Journal of the Chemical Society, Perkin Transactions 1; nb. 13; (2001); p. 1538 - 1549</p> <p><a href="#">View in Reaxys</a></p>

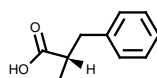

Rx-ID: 1522267 [View in Reaxys](#)

Yield	Conditions & References
52 %	<p><b>With</b> CuCl <b>in</b> tetrahydrofuran, Time= 0.25h, T= 0 °C</p> <p><b>Sato, Toshio; Kawara, Tatsuo; Kawashima, Masatoshi; Fujisawa, Tamotsu;</b> Chemistry Letters; (1980); p. 571 - 574</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2063579 [View in Reaxys](#)

Yield	Conditions & References
80 %	<p><b>With</b> KMnO<sub>4</sub></p> <p><b>Nalesnik, Theodore E.; Freudenberger, John H.; Orchin, Milton;</b> Journal of Organometallic Chemistry; <b>vol.</b> 221; nb. 2; (1981); p. 193 - 198</p> <p><a href="#">View in Reaxys</a></p>

(+)-2-methyl-3-phenyl-propionic acid


Rx-ID: 7672198 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> quinine</p> <p><b>Kenyon, Ross;</b> Journal of the Chemical Society; (1951); p. 3407,3409</p> <p><a href="#">View in Reaxys</a></p>

**Akabori; Sakurai;** Nippon Kagaku Zasshi; **vol.** 78; (1957); p. 1629; Chem.Abstr.; (1959); p. 21687

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**DeTar; Weis;** Journal of the American Chemical Society; **vol.** 79; (1957); p. 3045,3048

[View in Reaxys](#)

**Nerdel; John;** Chemische Berichte; **vol.** 89; (1956); p. 1945,1949

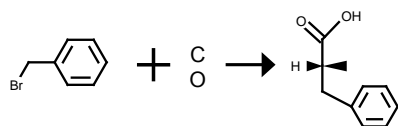
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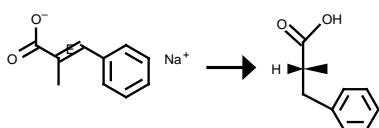
Rx-ID: 11430404 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1.1: 36 percent / LiHMDS / tetrahydrofuran / 12 h / -78 - 20 °C</p> <p>2.1: KHMDS / tetrahydrofuran; toluene / 1 h / -78 °C</p> <p>2.2: 38 percent / tetrahydrofuran; toluene / -78 - -40 °C</p> <p>3.1: 10 mg / aq. H<sub>2</sub>O<sub>2</sub>; LiOH / tetrahydrofuran / 16 h / 0 - 20 °C</p> <p><b>With</b> LiHMDS, KHMDS, aq. H<sub>2</sub>O<sub>2</sub>, LiOH <b>in</b> tetrahydrofuran, toluene</p> <p><b>Davies, Stephen G.; Garner, A. Christopher; Roberts, Paul M.; Smith, Andrew D.; Sweet, Miles J.; Thomson, James E.;</b> Organic and Biomolecular Chemistry; <b>vol.</b> 4; nb. 14; (2006); p. 2753 - 2768</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 3</p> <p>1.1: nBuLi / tetrahydrofuran / -78 °C</p> <p>1.2: 83 percent / tetrahydrofuran / -78 °C</p> <p>2.1: LDA / tetrahydrofuran / -78 °C</p> <p>2.2: 63 percent / tetrahydrofuran / -78 - 20 °C</p> <p>3.1: 78 percent / aq. LiOH / tetrahydrofuran / 0 - 20 °C</p> <p><b>With</b> nBuLi, LDA, aq. LiOH <b>in</b> tetrahydrofuran</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki;</b> Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 4</p> <p>1: 82 percent / Et<sub>3</sub>N / tetrahydrofuran / 1 h / 0 °C</p> <p>2: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>3: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>4: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Et<sub>3</sub>N, Mg, LDA, 3 M aq. HCl <b>in</b> tetrahydrofuran, diethyl ether, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjørn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 4</p> <p>1: 82 percent / Et<sub>3</sub>N / tetrahydrofuran / 1 h / 0 °C</p> <p>2: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>3: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>4: 76 percent / 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Et<sub>3</sub>N, Mg, LDA, 3 M aq. HCl <b>in</b> tetrahydrofuran, diethyl ether, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjørn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>

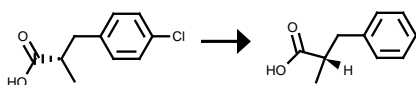



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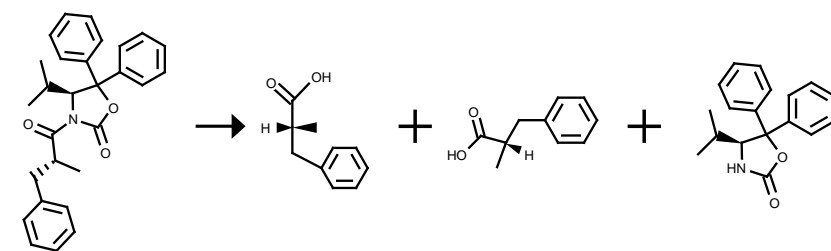
Yield	Conditions & References
	Reaction Steps: 2 1: 1.) LDA, LiCl / 1.) THF, from -78 to 23 deg C, 80 min, 2.) THF, 0 deg C, 0.2 h 2: 87 percent / 18 N H <sub>2</sub> SO <sub>4</sub> / dioxane / Heating <b>With</b> 1.) LDA, LiCl, 18 N H <sub>2</sub> SO <sub>4</sub> in dioxane  <b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 116; nb. 20; (1994); p. 9361 - 9362 <a href="#">View in Reaxys</a>


 Rx-ID: 3006378 [View in Reaxys](#)

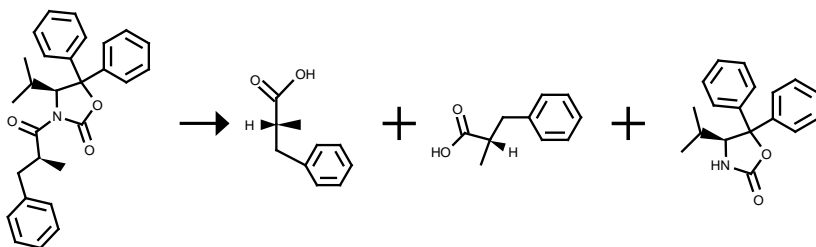
Yield	Conditions & References
95 %	<b>With</b> EDTA, phosphate buffer, Time= 12h, enoat-reductase  <b>Simon, Helmut; Guenther, Helmut; Bader, Johann; Tischer, Wilhelm;</b> Angewandte Chemie; <b>vol.</b> 93; nb. 10; (1981); p. 897 - 898 <a href="#">View in Reaxys</a>


 Rx-ID: 4807950 [View in Reaxys](#)

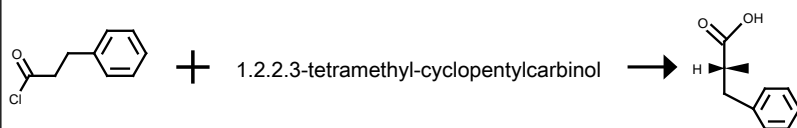
Yield	Conditions & References
	<b>With</b> H <sub>2</sub> , 10percent Pd/C in methanol, Time= 12h, Ambient temperature, Yield given  <b>Ferorelli, S.; Loidice, F.; Tortorella, V.; Amoroso, R.; Bettoni, G.; et al.;</b> Farmaco; <b>vol.</b> 52; nb. 6/7; (1997); p. 367 - 374 <a href="#">View in Reaxys</a>


 Rx-ID: 5041597 [View in Reaxys](#)

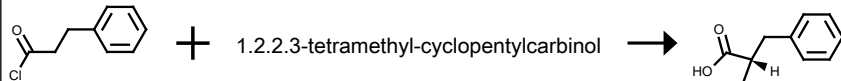
Yield	Conditions & References
94 %	<b>With</b> LiOH, Yield given  <b>Gibson, Colin L.; Gillon, Karen; Cook, Stuart;</b> Tetrahedron Letters; <b>vol.</b> 39; nb. 37; (1998); p. 6733 - 6736 <a href="#">View in Reaxys</a>


Rx-ID: 5041598 [View in Reaxys](#)

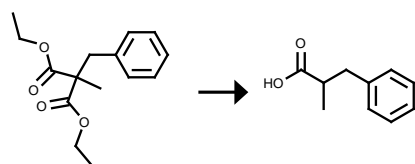
Yield	Conditions & References
95 %	<p><b>With</b> LiOH, Yield given</p> <p><b>Gibson, Colin L.; Gillon, Karen; Cook, Stuart;</b> Tetrahedron Letters; <b>vol.</b> 39; nb. 37; (1998); p. 6733 - 6736  <a href="#">View in Reaxys</a></p>


Rx-ID: 17172142 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 83 percent / Et<sub>3</sub>N / tetrahydrofuran / 0.17 h</p> <p>2: 95 percent / LiCl, diisopropylamine, n-BuLi / tetrahydrofuran; hexane / -78 °C</p> <p>3: aq. n-Bu<sub>4</sub>NOH / 2-methyl-propan-2-ol / Heating</p> <p><b>With</b> Et<sub>3</sub>N, LiCl, diisopropylamine, n-BuLi, aq. n-Bu<sub>4</sub>NOH in tetrahydrofuran, hexane, 2-methyl-propan-2-ol</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinsty, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>

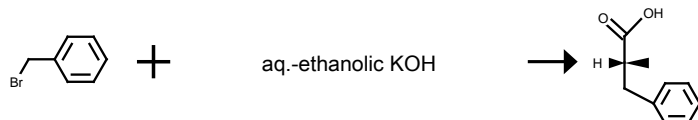

Rx-ID: 17172143 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 83 percent / Et<sub>3</sub>N / tetrahydrofuran / 0.17 h</p> <p>2: 95 percent / LiCl, diisopropylamine, n-BuLi / tetrahydrofuran; hexane / -78 °C</p> <p>3: aq. n-Bu<sub>4</sub>NOH / 2-methyl-propan-2-ol / Heating</p> <p><b>With</b> Et<sub>3</sub>N, LiCl, diisopropylamine, n-BuLi, aq. n-Bu<sub>4</sub>NOH in tetrahydrofuran, hexane, 2-methyl-propan-2-ol</p> <p><b>Myers, Andrew G.; Yang, Bryant H.; Chen, Hou; McKinsty, Lydia; Kopecky, David J.; Gleason, James L.;</b> Journal of the American Chemical Society; <b>vol.</b> 119; nb. 28; (1997); p. 6496 - 6511  <a href="#">View in Reaxys</a></p>

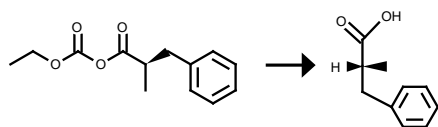

Rx-ID: 17763424 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p>

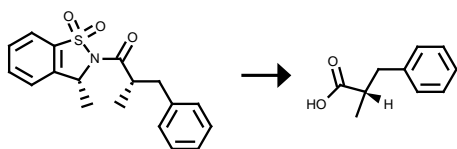
	<p>1: aq. KOH / 1 h / Heating  2: conc. H<sub>2</sub>SO<sub>4</sub> / 1 h / Heating  <b>With</b> aq. KOH, conc. H<sub>2</sub>SO<sub>4</sub></p> <p><b>Tyrrell, Elizabeth; Tsang, Michael W. H.; Skinner, George A.; Fawcett, John;</b> Tetrahedron; <b>vol.</b> 52; nb. 29; (1996); p. 9841 - 9852  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2  1: aq. KOH / ethanol / 24 h / Heating  2: dimethylformamide / 4 h / 100 - 110 °C  <b>With</b> aq. KOH in ethanol, dimethylformamide</p> <p><b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.;</b> Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2  1: concentrated aqueous KOH-solution  <b>With</b> concentrated aqueous KOH-solution</p> <p><b>Conrad; Bischoff;</b> Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 178  <a href="#">View in Reaxys</a></p>


Rx-ID: 19455142 [View in Reaxys](#)

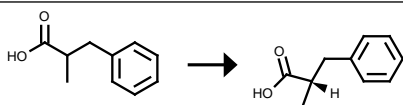
Yield	Conditions & References
	<p>Reaction Steps: 3  1: 1.) LiN(i-C<sub>3</sub>H<sub>7</sub>)<sub>2</sub> / 1.) THF, -78 deg C, 2.) 0 deg C, 4 h  2: 93 percent / PhCH<sub>2</sub>OH / tetrahydrofuran / 1 h / 0 °C  3: hydrogen  <b>With</b> 1.) LiN(i-C<sub>3</sub>H<sub>7</sub>)<sub>2</sub>, PhCH<sub>2</sub>OH, hydrogen in tetrahydrofuran</p> <p><b>Evans, D. A.; Ennis, M. D.; Mathre, D. J.;</b> Journal of the American Chemical Society; <b>vol.</b> 104; nb. 6; (1982); p. 1737 - 1739  <a href="#">View in Reaxys</a></p>


Rx-ID: 19468132 [View in Reaxys](#)

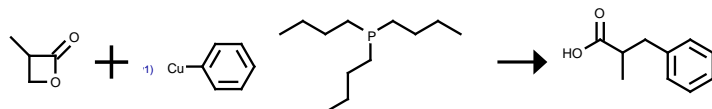
Yield	Conditions & References
	<p>Reaction Steps: 3  2: 93 percent / PhCH<sub>2</sub>OH / tetrahydrofuran / 1 h / 0 °C  3: hydrogen  <b>With</b> PhCH<sub>2</sub>OH, hydrogen in tetrahydrofuran</p> <p><b>Evans, D. A.; Ennis, M. D.; Mathre, D. J.;</b> Journal of the American Chemical Society; <b>vol.</b> 104; nb. 6; (1982); p. 1737 - 1739  <a href="#">View in Reaxys</a></p>


Rx-ID: 19572904 [View in Reaxys](#)

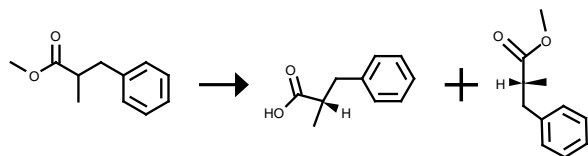
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 86 percent / Ti(i-PrO)<sub>4</sub>, molecular sieves / 3 h / 150 °C</p> <p>2: 89 percent / H<sub>2</sub>O / &lt;RhCl(PPh<sub>3</sub>)<sub>3</sub>&gt; / ethanol / 5 h / Heating</p> <p><b>With</b> Ti(i-PrO)<sub>4</sub>, molecular sieves, H<sub>2</sub>O, &lt;RhCl(PPh<sub>3</sub>)<sub>3</sub>&gt; <b>in</b> ethanol</p> <p><b>Oppolzer, Wolfgang; Lienard, Philippe</b>; Helvetica Chimica Acta; <b>vol.</b> 75; nb. 8; (1992); p. 2572 - 2582</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 438148 [View in Reaxys](#)

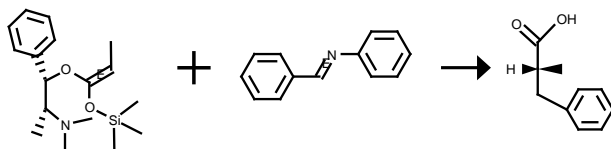
Yield	Conditions & References
	<p><b>With</b> l-menthylamine</p> <p><b>Pickard; Yates</b>; Journal of the Chemical Society; <b>vol.</b> 95; (1909); p. 1018</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> quinine</p> <p><b>Kipping; Hunter</b>; Journal of the Chemical Society; <b>vol.</b> 83; (1903); p. 1007</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>Cram,D.J.; Wingrove,A.S.</b>; Journal of the American Chemical Society; <b>vol.</b> 86; (1964); p. 5490 - 5496</p> <p><a href="#">View in Reaxys</a></p> <p><b>Watson,M.B.; Youngson,G.W.</b>; Journal of the Chemical Society [Section] C: Organic; (1968); p. 258 - 262</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2</p> <p>1: HCl / 24 h / Ambient temperature</p> <p>2: water / 2 h / lipase PS</p> <p><b>With</b> HCl, water</p> <p><b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.</b>; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 1522269 [View in Reaxys](#)

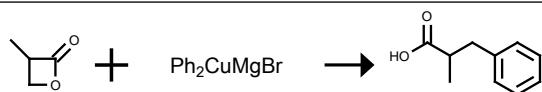
Yield	Conditions & References
67 %	<p><b>in</b> diethyl ether, Time= 1.5h, T= -78 °C</p> <p><b>Kawashima, Masatoshi; Sato, Toshio; Fujisawa, Tamotsu</b>; Tetrahedron; <b>vol.</b> 45; nb. 2; (1989); p. 403 - 412</p> <p><a href="#">View in Reaxys</a></p>


 Rx-ID: 2042020 [View in Reaxys](#)

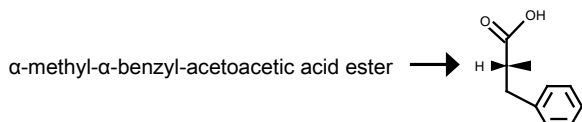
Yield	Conditions & References
44 %, 50 %	<b>With</b> lipase from <i>Pseudomonas</i> sp. (Amano), Time= 3h, Ambient temperature, 0.1 phosphate buffer, pH 7.0 <b>Delnick, Deborah L.; Margolin, Alexey L.</b> ; Tetrahedron Letters; <b>vol.</b> 31; nb. 47; (1990); p. 6797 - 6798 <a href="#">View in Reaxys</a>
50 %, 44 %	<b>With</b> lipase from <i>Pseudomonas</i> sp. (Amano), Time= 3h, Ambient temperature, 0.1 phosphate buffer, pH 7.0 <b>Delnick, Deborah L.; Margolin, Alexey L.</b> ; Tetrahedron Letters; <b>vol.</b> 31; nb. 47; (1990); p. 6797 - 6798 <a href="#">View in Reaxys</a>


 Rx-ID: 3760449 [View in Reaxys](#)

Yield	Conditions & References
	<b>With</b> 1) TiCl <sub>4</sub> , 2) H <sub>2</sub> , 2) PdCl <sub>2</sub> , 1) CH <sub>2</sub> Cl <sub>2</sub> , -78 deg C, 2 h, then -40 deg C, 2 h, 2) CH <sub>3</sub> COOH, Multistep reaction <b>Gennari, Cesare; Venturini, Isabella; Gislou, Gabriele; Schimperna, Giuliana</b> ; Tetrahedron Letters; <b>vol.</b> 28; nb. 2; (1987); p. 227 - 230 <a href="#">View in Reaxys</a>
	<b>With</b> 1.) TiCl <sub>4</sub> ; 2.) H <sub>2</sub> , 2.) PdCl <sub>2</sub> , 1.) CH <sub>2</sub> Cl <sub>2</sub> , -78 - -40 deg C, 3 h; 2.) AcOH, 5 d, Yield given. Multistep reaction <b>Gennari, Cesare; Schimperna, Giuliana; Venturini, Isabella</b> ; Tetrahedron; <b>vol.</b> 44; nb. 13; (1988); p. 4221 - 4232 <a href="#">View in Reaxys</a>


 Rx-ID: 7250401 [View in Reaxys](#)

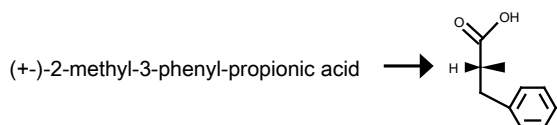
Yield	Conditions & References
52 %	<b>in</b> tetrahydrofuran, various solvent(s), 1.) -30 deg C, 1 h, 2.) 0 deg C, 1 h <b>Fujisawa, Tamotsu; Sato, Toshio; Kawara, Tatsuo; Kawashima, Masatoshi</b> ; Tetrahedron Letters; <b>vol.</b> 21; (1980); p. 2181 - 2184 <a href="#">View in Reaxys</a>


 Rx-ID: 7625856 [View in Reaxys](#)

Yield	Conditions & References
	<b>With</b> KOH-solution

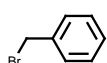
Jones; Wallis; Journal of the American Chemical Society; **vol.** 48; (1926); p. 180

[View in Reaxys](#)

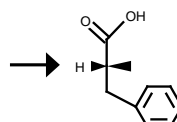


Rx-ID: 7625857 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> quinine</p> <p><b>DeTar; Weis;</b> Journal of the American Chemical Society; <b>vol.</b> 79; (1957); p. 3045,3048</p> <p><a href="#">View in Reaxys</a></p>

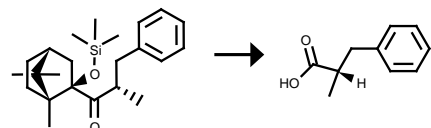


polymer-bound  $\text{NMe}_3^{(1+)}\text{SCN}^{(1-)}$



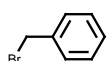
Rx-ID: 15262192 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1.1: <math>\text{KN}(\text{SiMe}_3)_2</math> / tetrahydrofuran; toluene / 4 h / -78 °C</p> <p>1.2: 80 percent / tetrahydrofuran; toluene / -78 - 50 °C</p> <p>2.1: 95 percent / CAN / acetonitrile; <math>\text{H}_2\text{O}</math> / 1 h / 0 °C</p> <p><b>With</b> <math>\text{KN}(\text{SiMe}_3)_2</math>, CAN in tetrahydrofuran, toluene, acetonitrile, <math>\text{H}_2\text{O}</math></p> <p><b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Le-cumberri, Ainara; Linden, Anthony;</b> Organic Letters; <b>vol.</b> 3; nb. 21; (2001); p. 3249 - 3252</p> <p><a href="#">View in Reaxys</a></p>

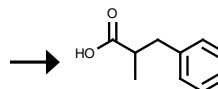


Rx-ID: 15284627 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 85 percent / TBAF / tetrahydrofuran / 0.5 h / 20 °C</p> <p>2: 90 percent / CAN / acetonitrile; <math>\text{H}_2\text{O}</math> / 1 h / 0 °C</p> <p><b>With</b> TBAF, CAN in tetrahydrofuran, acetonitrile, <math>\text{H}_2\text{O}</math></p> <p><b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Le-cumberri, Ainara; Linden, Anthony;</b> Organic Letters; <b>vol.</b> 3; nb. 21; (2001); p. 3249 - 3252</p> <p><a href="#">View in Reaxys</a></p>



polymer-bound  $\text{NMe}_3^{(1+)}\text{SCN}^{(1-)}$



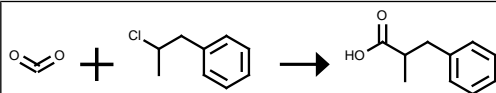
Rx-ID: 15507334 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: LDA / tetrahydrofuran</p> <p>2: KOH; <math>\text{H}_2\text{O}</math></p>

With LDA, KOH, H<sub>2</sub>O in tetrahydrofuran

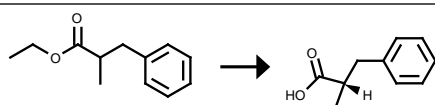
**Kane, John L.; Shea, Kevin M.; Crombie, Aimee L.; Danheiser, Rick L.**; Organic Letters; **vol.** 3; nb. 7; (2001); p. 1081 - 1084

[View in Reaxys](#)



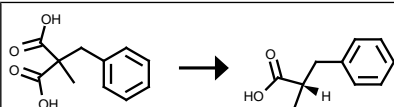
Rx-ID: 1046610 [View in Reaxys](#)

Yield	Conditions & References
	With Mg in diethyl ether
	Urry, W.H. et al.; Journal of Organic Chemistry; <b>vol.</b> 29; nb. 7; (1964); p. 1663 - 1669
	<a href="#">View in Reaxys</a>



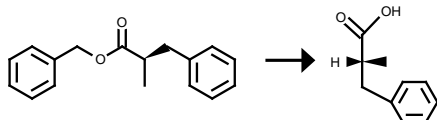
Rx-ID: 1156924 [View in Reaxys](#)

Yield	Conditions & References
	With α-chymotrypsin, aq. NaOH
	Cohen, S.G.; Milovanovic, A.; Journal of the American Chemical Society; <b>vol.</b> 90; (1968); p. 3495 - 3502
	<a href="#">View in Reaxys</a>



Rx-ID: 1364727 [View in Reaxys](#)

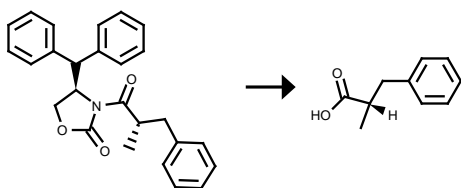
Yield	Conditions & References
	(decarboxylation)
	Terashima, S. et al.; Chemical and Pharmaceutical Bulletin; <b>vol.</b> 18; (1970); p. 1124 - 1136
	<a href="#">View in Reaxys</a>
	Reaction Steps: 3
	1: dimethylformamide / 4 h / 100 - 110 °C
	2: HCl / 24 h / Ambient temperature
	3: water / 2 h / lipase PS
	With HCl, water in dimethylformamide
	Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030
	<a href="#">View in Reaxys</a>



Rx-ID: 2774511 [View in Reaxys](#)

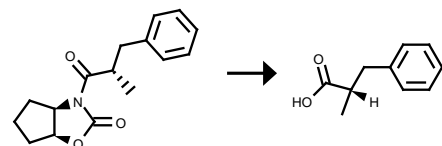
Yield	Conditions & References
	With hydrogen

**Evans, D. A.; Ennis, M. D.; Mathre, D. J.**; Journal of the American Chemical Society; **vol.** 104; nb. 6; (1982); p. 1737 - 1739  
[View in Reaxys](#)



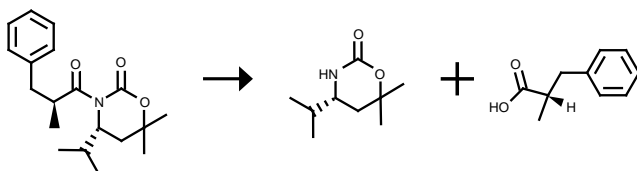
Rx-ID: 4369811 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> LiO<sub>2</sub>H in tetrahydrofuran, H<sub>2</sub>O, T= 0 °C</p> <p><b>Sibi, Mukund P.; Deshpande, Prasad K.; Ji, Jianguo</b>; Tetrahedron Letters; <b>vol.</b> 36; nb. 49; (1995); p. 8965 - 8968  <a href="#">View in Reaxys</a></p>



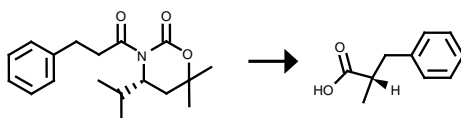
Rx-ID: 4699524 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> aq. LiOH, H<sub>2</sub>O<sub>2</sub> in tetrahydrofuran, T= 0 °C</p> <p><b>Ghosh, Arun K.; Cho, Hanna; Onishi, Masanobu</b>; Tetrahedron: Asymmetry; <b>vol.</b> 8; nb. 6; (1997); p. 821 - 824  <a href="#">View in Reaxys</a></p>



Rx-ID: 11137429 [View in Reaxys](#)

Yield	Conditions & References
11 mg, 10 mg	<p><b>With</b> aq. H<sub>2</sub>O<sub>2</sub>, LiOH in tetrahydrofuran, Time= 16h, T= 0 - 20 °C</p> <p><b>Davies, Stephen G.; Garner, A. Christopher; Roberts, Paul M.; Smith, Andrew D.; Sweet, Miles J.; Thomson, James E.</b>; Organic and Biomolecular Chemistry; <b>vol.</b> 4; nb. 14; (2006); p. 2753 - 2768  <a href="#">View in Reaxys</a></p>



Rx-ID: 11400860 [View in Reaxys](#)

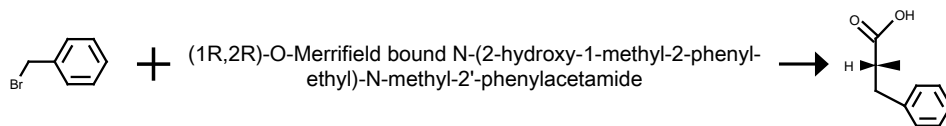
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1.1: KHMDS / tetrahydrofuran; toluene / 1 h / -78 °C</p> <p>1.2: 38 percent / tetrahydrofuran; toluene / -78 - -40 °C</p> <p>2.1: 10 mg / aq. H<sub>2</sub>O<sub>2</sub>; LiOH / tetrahydrofuran / 16 h / 0 - 20 °C</p>



With KHMDS, aq. H<sub>2</sub>O<sub>2</sub>, LiOH in tetrahydrofuran, toluene

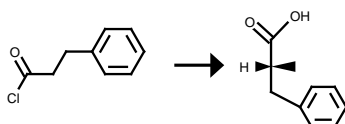
**Davies, Stephen G.; Garner, A. Christopher; Roberts, Paul M.; Smith, Andrew D.; Sweet, Miles J.; Thomson, James E.**; Organic and Biomolecular Chemistry; **vol.** 4; nb. 14; (2006); p. 2753 - 2768

[View in Reaxys](#)



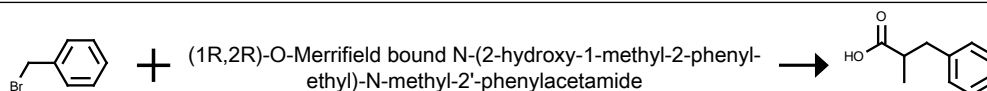
Rx-ID: 11423706 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1.1: KHMDS / tetrahydrofuran; toluene / 1 h / -78 °C</p> <p>1.2: 78 percent / 5 h / -40 °C</p> <p>2.1: 100 percent / aq. H<sub>2</sub>O<sub>2</sub>; LiOH / tetrahydrofuran / 16 h / 0 - 20 °C</p> <p><b>With</b> KHMDS, aq. H<sub>2</sub>O<sub>2</sub>, LiOH <b>in</b> tetrahydrofuran, toluene</p> <p><b>Davies, Stephen G.; Garner, A. Christopher; Roberts, Paul M.; Smith, Andrew D.; Sweet, Miles J.; Thomson, James E.</b>; Organic and Biomolecular Chemistry; <b>vol.</b> 4; nb. 14; (2006); p. 2753 - 2768</p> <p><a href="#">View in Reaxys</a></p>



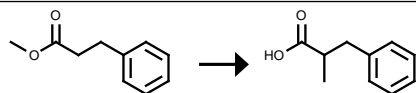
Rx-ID: 11786208 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1.1: nBuLi / tetrahydrofuran / -78 °C</p> <p>1.2: 83 percent / tetrahydrofuran / -78 °C</p> <p>2.1: LDA / tetrahydrofuran / -78 °C</p> <p>2.2: 63 percent / tetrahydrofuran / -78 - 20 °C</p> <p>3.1: 77 percent / aq. LiOH / tetrahydrofuran / 0 - 20 °C</p> <p><b>With</b> nBuLi, LDA, aq. LiOH <b>in</b> tetrahydrofuran</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki</b>; Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 4</p> <p>1: 82 percent / Et<sub>3</sub>N / tetrahydrofuran / 1 h / 0 °C</p> <p>2: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>3: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>4: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Et<sub>3</sub>N, Mg, LDA, 3 M aq. HCl <b>in</b> tetrahydrofuran, diethyl ether, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjørn</b>; Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>

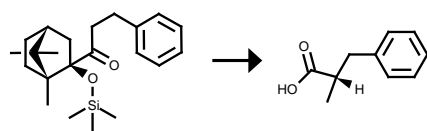


Rx-ID: 13176314 [View in Reaxys](#)

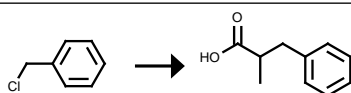
Yield	Conditions & References
	Reaction Steps: 2 1: LDA / tetrahydrofuran 2: aq. KOH <b>With</b> LDA, aq. KOH in tetrahydrofuran  <b>Crombie, Aimee L.; Kane, John L.; Shea, Kevin M.; Danheiser, Rick L.</b> ; Journal of Organic Chemistry; <b>vol.</b> 69; nb. 25; (2004); p. 8652 - 8667 <a href="#">View in Reaxys</a>


 Rx-ID: 14077256 [View in Reaxys](#)

Yield	Conditions & References
	Reaction Steps: 2 1.1: diisopropylamine; n-butyllithium / tetrahydrofuran; hexane / 0.33 h / -78 °C 1.2: 47 percent / tetrahydrofuran; hexane / 0.5 h / 0 °C 2.1: 100 percent / potassium hydroxide / ethanol; H <sub>2</sub> O / 8 h / 0 °C <b>With</b> diisopropylamine, n-butyllithium, potassium hydroxide in tetrahydrofuran, hexane, ethanol, H <sub>2</sub> O  <b>Kato, Dai-ichiro; Mitsuda, Satoshi; Ohta, Hiromichi</b> ; Journal of Organic Chemistry; <b>vol.</b> 68; nb. 19; (2003); p. 7234 - 7242 <a href="#">View in Reaxys</a>


 Rx-ID: 15284474 [View in Reaxys](#)

Yield	Conditions & References
	Reaction Steps: 3 1.1: LDA; DMPU / tetrahydrofuran; hexane / 4 h / -50 °C 1.2: tetrahydrofuran; hexane / 1 h / -30 °C 2.1: 85 percent / TBAF / tetrahydrofuran / 0.5 h / 20 °C 3.1: 90 percent / CAN / acetonitrile; H <sub>2</sub> O / 1 h / 0 °C <b>With</b> LDA, DMPU, TBAF, CAN in tetrahydrofuran, hexane, acetonitrile, H <sub>2</sub> O  <b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Le-cumberri, Ainara; Linden, Anthony</b> ; Organic Letters; <b>vol.</b> 3; nb. 21; (2001); p. 3249 - 3252 <a href="#">View in Reaxys</a>


 Rx-ID: 21039728 [View in Reaxys](#)

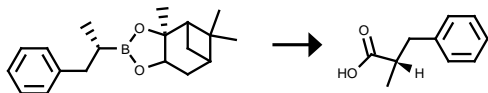
Yield	Conditions & References
	Reaction Steps: 3 1: 1.) NaH / 1.) ethyl ether, reflux, 2.) ethyl ether, RT, 6 h 2: aq. KOH / ethanol / 24 h / Heating 3: dimethylformamide / 4 h / 100 - 110 °C <b>With</b> 1.) NaH, aq. KOH in ethanol, dimethylformamide  <b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.</b> ; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030

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Reaction Steps: 3  
2: concentrated aqueous KOH-solution  
**With** concentrated aqueous KOH-solution

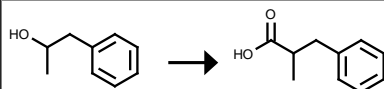
**Conrad; Bischoff;** Justus Liebigs Annalen der Chemie; **vol.** 204; (1880); p. 178

[View in Reaxys](#)



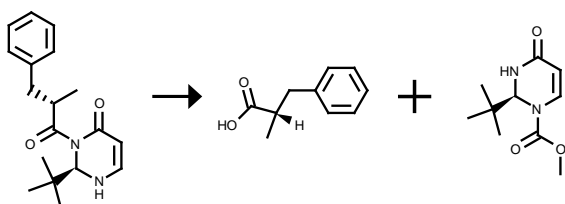
Rx-ID: 21089128 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2 2: 90 percent / chromic acid, water / diethyl ether / 2 h / T &lt;25 deg C <b>With</b> chromic acid, water in diethyl ether</p> <p><b>Rangaishenvi, Milind V.; Singaram, Bakthan; Brown, Herbert C.;</b> Journal of Organic Chemistry; <b>vol.</b> 56; nb. 10; (1991); p. 3286 - 3294</p> <p><a href="#">View in Reaxys</a></p>



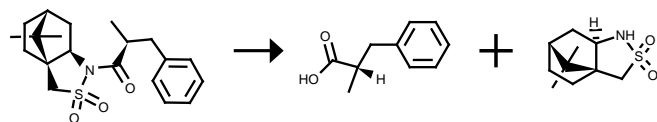
Rx-ID: 22518041 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2 1: SOCl<sub>2</sub> 2: Mg / diethyl ether <b>With</b> SOCl<sub>2</sub>, Mg in diethyl ether</p> <p><b>Urry, W.H. et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 29; nb. 7; (1964); p. 1663 - 1669</p> <p><a href="#">View in Reaxys</a></p>

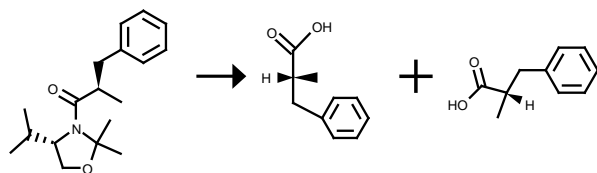


Rx-ID: 2510307 [View in Reaxys](#)

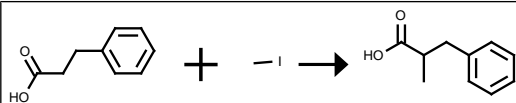
Yield	Conditions & References
	<p><b>With</b> 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 2.) methyl chloroformate, 1.) THF, 30 min, 2.) THF, RT, 4 h, Yield given. Multistep reaction. Yields of byproduct given</p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.;</b> Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2691104 [View in Reaxys](#)

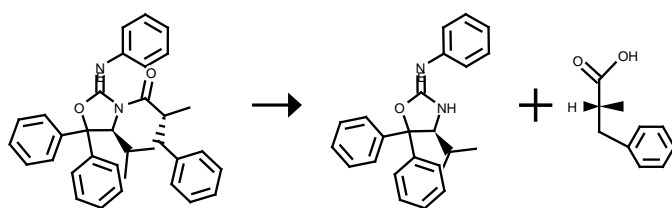
Yield	Conditions & References
	<p><b>With</b> H<sub>2</sub>O<sub>2</sub> (30percent), LiOH·H<sub>2</sub>O in tetrahydrofuran, H<sub>2</sub>O, 1.) 0 deg C, 1 h; 2.) room temp., 16 h</p> <p><b>Oppolzer, Wolfgang; Moretti, Robert; Thomi, Silvia;</b> Tetrahedron Letters; <b>vol.</b> 30; nb. 41; (1989); p. 5603 - 5606</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 4289289 [View in Reaxys](#)

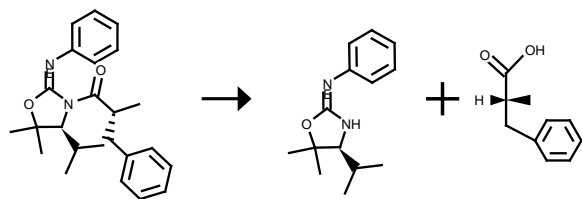
Yield	Conditions & References
	<p><b>With</b> 6 N sulfuric acid in acetic acid, Time= 4h, Heating, Yield given. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Kanemasa, Shuji; Ueno, Koichi; Onimura, Kenjiro; Kikukawa, Takashi; Yamamoto, Hidetoshi;</b> Tetrahedron; <b>vol.</b> 51; nb. 38; (1995); p. 10453 - 10462</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11048898 [View in Reaxys](#)

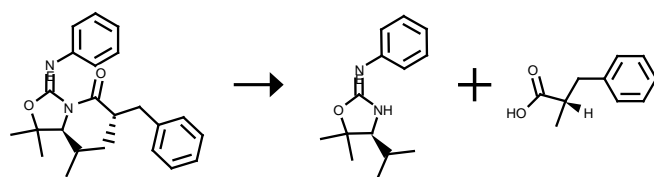
Yield	Conditions & References
	<p><b>With</b> LDA in tetrahydrofuran, T= -78 °C</p> <p><b>Zhang, Minsheng; Porte, Alex; Diamantidis, George; Sogi, Kimberly; Kubrak, Dennis; Resnick, Lynn; Mayer, Scott C.; Wang, Zheng; Kreft, Anthony F.; Harrison, Boyd L.;</b> Bioorganic and Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 9; (2007); p. 2401 - 2403</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11198115 [View in Reaxys](#)

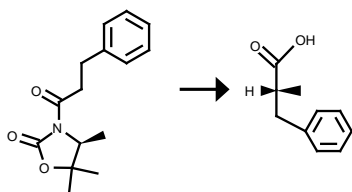
Yield	Conditions & References
	<p><b>With</b> aq. NaOH in dioxane, Time= 1h, Heating</p> <p><b>Le, Thanh Nguyen; Nguyen, Quynh Pham Bao; Kim, Jae Nyoung; Kim, Taek Hyeon;</b> Tetrahedron Letters; <b>vol.</b> 48; nb. 44; (2007); p. 7834 - 7837</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11198117 [View in Reaxys](#)

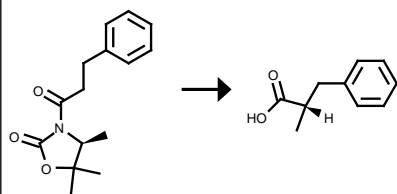
Yield	Conditions & References
	<p><b>With</b> aq. NaOH in dioxane, Time= 1h, Heating</p> <p><b>Le, Thanh Nguyen; Nguyen, Quynh Pham Bao; Kim, Jae Nyoung; Kim, Taek Hyeon;</b> Tetrahedron Letters; <b>vol.</b> 48; nb. 44; (2007); p. 7834 - 7837</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11198119 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> aq. NaOH in dioxane, Time= 1h, Heating</p> <p><b>Le, Thanh Nguyen; Nguyen, Quynh Pham Bao; Kim, Jae Nyoung; Kim, Taek Hyeon;</b> Tetrahedron Letters; <b>vol.</b> 48; nb. 44; (2007); p. 7834 - 7837</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11785689 [View in Reaxys](#)

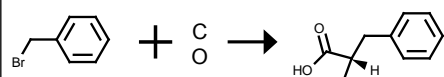
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1.1: LDA / tetrahydrofuran / -78 °C</p> <p>1.2: 63 percent / tetrahydrofuran / -78 - 20 °C</p> <p>2.1: 77 percent / aq. LiOH / tetrahydrofuran / 0 - 20 °C</p> <p><b>With</b> LDA, aq. LiOH in tetrahydrofuran</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki;</b> Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 11785690 [View in Reaxys](#)

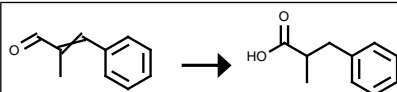
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1.1: LDA / tetrahydrofuran / -78 °C</p> <p>1.2: 63 percent / tetrahydrofuran / -78 - 20 °C</p> <p>2.1: 78 percent / aq. LiOH / tetrahydrofuran / 0 - 20 °C</p> <p><b>With</b> LDA, aq. LiOH in tetrahydrofuran</p> <p><b>Suzuki, Takayoshi; Hisakawa, Shinya; Itoh, Yukihiro; Maruyama, Sakiko; Kurotaki, Mineko; Nakagawa, Hidehiko; Miyata, Naoki;</b> Bioorganic &amp; Medicinal Chemistry Letters; <b>vol.</b> 17; nb. 6; (2007); p. 1558 - 1561</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 15266847 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 4</p> <p>1.1: LDA / tetrahydrofuran; hexane / 2 h / -78 °C</p> <p>1.2: 80 percent / tetrahydrofuran; hexane / 1.5 h / -30 °C</p> <p>2.1: LDA; DMPU / tetrahydrofuran; hexane / 4 h / -50 °C</p> <p>2.2: tetrahydrofuran; hexane / 1 h / -30 °C</p> <p>3.1: 85 percent / TBAF / tetrahydrofuran / 0.5 h / 20 °C</p> <p>4.1: 90 percent / CAN / acetonitrile; H<sub>2</sub>O / 1 h / 0 °C</p> <p><b>With</b> LDA, DMPU, TBAF, CAN in tetrahydrofuran, hexane, acetonitrile, H<sub>2</sub>O</p> <p><b>Palomo, Claudio; Oiarbide, Mikel; Mielgo, Antonia; Gonzalez, Alberto; Garcia, Jesus M.; Landa, Cristina; Le-cumberri, Ainara; Linden, Anthony;</b> Organic Letters; <b>vol.</b> 3; nb. 21; (2001); p. 3249 - 3252</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 17971757 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 81 percent / NaHMDS / tetrahydrofuran / -78 to 0 deg C</p> <p>2: LiO<sub>2</sub>H / tetrahydrofuran; H<sub>2</sub>O / 0 °C</p> <p><b>With</b> NaHMDS, LiO<sub>2</sub>H in tetrahydrofuran, H<sub>2</sub>O</p> <p><b>Sibi, Mukund P.; Deshpande, Prasad K.; Ji, Jianguo;</b> Tetrahedron Letters; <b>vol.</b> 36; nb. 49; (1995); p. 8965 - 8968</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 18527359 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 96 percent / NaOAc, PPh<sub>3</sub>, H<sub>2</sub>, RhCl(PPh<sub>3</sub>) / benzene / 24 h / 110 °C / 103430 Torr</p> <p>2: 80 percent / KMnO<sub>4</sub></p> <p><b>With</b> NaOAc, PPh<sub>3</sub>, H<sub>2</sub>, RhCl(PPh<sub>3</sub>), KMnO<sub>4</sub> <b>in</b> benzene</p> <p><b>Nalesnik, Theodore E.; Freudenberger, John H.; Orchin, Milton</b>; Journal of Organometallic Chemistry; <b>vol.</b> 221; nb. 2; (1981); p. 193 - 198</p> <p><a href="#">View in Reaxys</a></p>

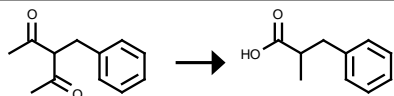


Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 89 percent / 4-(N,N-dimethylamino)pyridine / pyridine-d<sub>5</sub> / 0.05 h / 80 °C</p> <p>2: 89 percent / 3percent methanolic HCl / 16 h / Ambient temperature</p> <p><b>With</b> 4-(N,N-dimethylamino)pyridine, 3percent methanolic HCl <b>in</b> pyridine-d<sub>5</sub></p> <p><b>Wilson, Kenneth E.; Burk, Robert M.; Biftu, Tesfaye; Ball, Richard G.; Hoogsteen, Karst</b>; Journal of Organic Chemistry; <b>vol.</b> 57; nb. 28; (1992); p. 7151 - 7158</p> <p><a href="#">View in Reaxys</a></p>



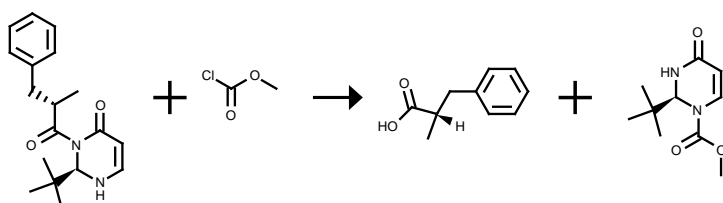
Yield	Conditions & References
	<p><b>With</b> chromic acid, acetic acid</p> <p><b>Guerbet</b>; Bulletin de la Societe Chimique de France; <b>vol.</b> &lt;4&gt; 3; (1908); p. 945; Chem. Zentralbl.; <b>vol.</b> 79; nb. II; (1908); p. 866</p> <p><a href="#">View in Reaxys</a></p> <p><b>Guerbet</b>; Comptes Rendus Hebdomadaires des Seances de l'Academie des Sciences; <b>vol.</b> 146; (1908); p. 1406; Bulletin de la Societe Chimique de France; <b>vol.</b> &lt;4&gt; 3; (1908); p. 944; Chem. Zentralbl.; <b>vol.</b> 79; nb. II; (1908); p. 866</p> <p><a href="#">View in Reaxys</a></p>
	<p><b>With</b> potash, T= 230 °C , im geschlossenen Rohr</p> <p><b>Guerbet</b>; Bulletin de la Societe Chimique de France; <b>vol.</b> &lt;4&gt; 3; (1908); p. 945; Chem. Zentralbl.; <b>vol.</b> 79; nb. II; (1908); p. 866</p> <p><a href="#">View in Reaxys</a></p> <p><b>Guerbet</b>; Comptes Rendus Hebdomadaires des Seances de l'Academie des Sciences; <b>vol.</b> 146; (1908); p. 1406; Bulletin de la Societe Chimique de France; <b>vol.</b> &lt;4&gt; 3; (1908); p. 944; Chem. Zentralbl.; <b>vol.</b> 79; nb. II; (1908); p. 866</p> <p><a href="#">View in Reaxys</a></p>
	<p>bei der Oxydation</p> <p><b>Guerbet</b>; Comptes Rendus Hebdomadaires des Seances de l'Academie des Sciences; <b>vol.</b> 146; (1908); p. 300,1405</p> <p><a href="#">View in Reaxys</a></p>

**Guerbet**; Bulletin de la Societe Chimique de France; **vol.** <4> 3; (1908); p. 504,943; Chem. Zentralbl.; **vol.** 79; nb. II; (1908); p. 866  
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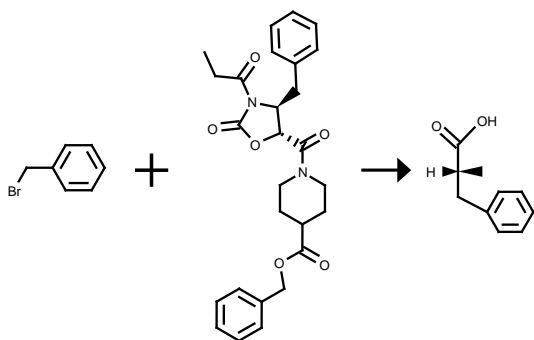
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Yield	Conditions & References
	<p><b>With</b> H<sub>2</sub>O<sub>2</sub>, aq. NaOH in methanol</p> <p><b>Cocker,W.; Grayson,D.H.</b>; Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); (1975); p. 1347 - 1352  <a href="#">View in Reaxys</a></p>



Rx-ID: 3356190 [View in Reaxys](#)

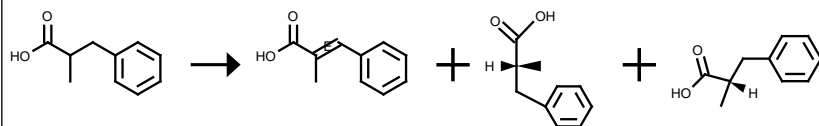
Yield	Conditions & References
	<p><b>With</b> 1.) 30percent aq. H<sub>2</sub>O<sub>2</sub>, aq. LiOH, 1.) THF, 30 min, 2.) THF, RT, 4 h, Yield given. Multistep reaction. Yields of byproduct given</p> <p><b>Chu, Kent S.; Negrete, George R.; Konopelski, Joseph P.; Lakner, Frederick J.; Woo, Nam-Tae; Olmstead, Marilyn M.</b>; Journal of the American Chemical Society; <b>vol.</b> 114; nb. 5; (1992); p. 1800 - 1812  <a href="#">View in Reaxys</a></p>



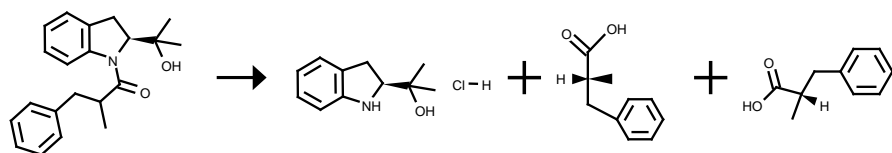
Rx-ID: 9658934 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>Stage 1: With</b> LDA in tetrahydrofuran, T= -78 - 0 °C</p> <p><b>Stage 2: With</b> LiOH, aq. H<sub>2</sub>O<sub>2</sub> in tetrahydrofuran, T= 0 °C , Further stages.</p> <p><b>Kotake, Tomoya; Rajesh, S.; Hayashi, Yoshio; Mukai, Yoshie; Ueda, Mitsuhiro; Kimura, Tooru; Kiso, Yoshia-ki</b>; Tetrahedron Letters; <b>vol.</b> 45; nb. 18; (2004); p. 3651 - 3654  <a href="#">View in Reaxys</a></p>

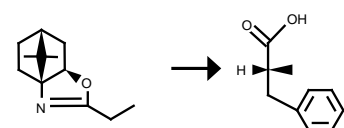



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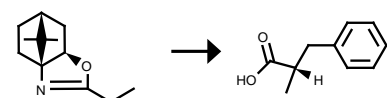
Yield	Conditions & References
	<p><b>With</b> <i>Nocardia diaphanozonaria</i> JCM 3208 whole cells, phosphate buffer, 2-bromohexanoic acid in H<sub>2</sub>O, T= 30 °C , pH= 7.0, Title compound not separated from byproducts</p> <p><b>Kato, Dai-ichiro; Miyamoto, Kenji; Ohta, Hiromichi;</b> <i>Tetrahedron: Asymmetry</i>; <b>vol.</b> 15; nb. 18; (2004); p. 2965 - 2974</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 10467878 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> aq. HCl in dioxane</p> <p><b>Andersson, Fredrik; Hedenstroem, Erik;</b> <i>Tetrahedron: Asymmetry</i>; <b>vol.</b> 17; nb. 13; (2006); p. 1952 - 1957</p> <p><a href="#">View in Reaxys</a></p>

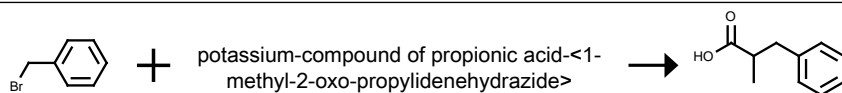

Rx-ID: 15735867 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1.1: LDA / tetrahydrofuran / -78 °C</p> <p>1.2: 95 percent / tetrahydrofuran</p> <p>2.1: 24 h / Heating</p> <p>3.1: KOH; MeOH / Heating</p> <p><b>With</b> LDA, KOH, MeOH in tetrahydrofuran, 1.1: Metallation / 1.2: Alkylation / 2.1: Ring cleavage / 3.1: Hydrolysis</p> <p><b>Chandrasekhar, Sosale; Kausar, Amina;</b> <i>Tetrahedron: Asymmetry</i>; <b>vol.</b> 11; nb. 11; (2000); p. 2249 - 2254</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 15735868 [View in Reaxys](#)

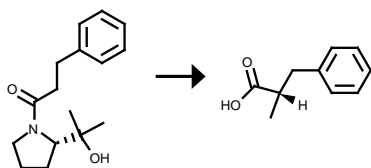
Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1.1: LDA / tetrahydrofuran / -78 °C</p> <p>1.2: 95 percent / tetrahydrofuran</p> <p>2.1: 24 h / Heating</p> <p>3.1: KOH; MeOH / Heating</p> <p><b>With</b> LDA, KOH, MeOH in tetrahydrofuran, 1.1: Metallation / 1.2: Alkylation / 2.1: Ring cleavage / 3.1: Hydrolysis</p>

**Chandrasekhar, Sosale; Kausar, Amina;** Tetrahedron: Asymmetry; **vol.** 11; nb. 11; (2000); p. 2249 - 2254  
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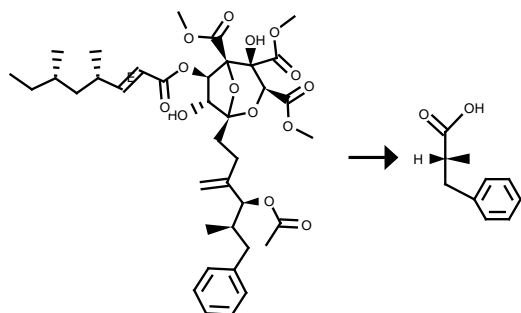
Rx-ID: 17508389 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 80 percent / LDA / tetrahydrofuran / 1.) 0 deg C, 45 min, 2.) r.t., 1 h</p> <p>2: 5 percent / manganese(III) acetate / 22 h / 60 °C</p> <p><b>With</b> LDA, manganese(III) acetate <b>in</b> tetrahydrofuran</p> <p><b>Jamie, Joanne F.; Rickards, Rodney W.;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 21; (1996); p. 2603 - 2614  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2</p> <p>1: 80 percent / LDA / tetrahydrofuran / 1.) 0 deg C, 45 min, 2.) r.t., 1 h</p> <p>2: 5 percent / manganese(III) acetate, acetic acid / 22 h / 60 °C</p> <p><b>With</b> LDA, manganese(III) acetate, acetic acid <b>in</b> tetrahydrofuran</p> <p><b>Jamie, Joanne F.; Rickards, Rodney W.;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 21; (1996); p. 2603 - 2614  <a href="#">View in Reaxys</a></p>

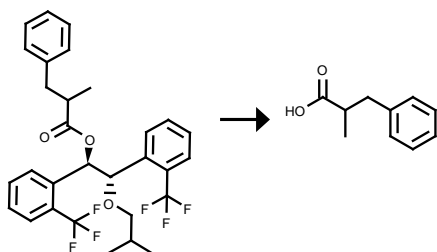


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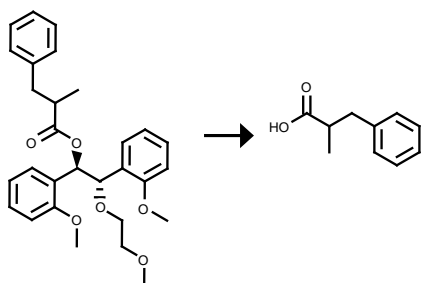
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>2: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> LDA, 3 M aq. HCl <b>in</b> tetrahydrofuran, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjorn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802  <a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 2</p> <p>1: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>2: 76 percent / 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> LDA, 3 M aq. HCl <b>in</b> tetrahydrofuran, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjorn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802  <a href="#">View in Reaxys</a></p>


Rx-ID: 19576306 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 4</p> <p>1: 68 percent / 0.2percent K<sub>2</sub>CO<sub>3</sub> / methanol / 1.33 h / 30 °C</p> <p>2: 89 percent / 4-(N,N-dimethylamino)pyridine / pyridine-d<sub>5</sub> / 0.05 h / 80 °C</p> <p>3: 89 percent / 3percent methanolic HCl / 16 h / Ambient temperature</p> <p><b>With</b> 0.2percent K<sub>2</sub>CO<sub>3</sub>, 4-(N,N-dimethylamino)pyridine, 3percent methanolic HCl <b>in</b> methanol, pyridine-d<sub>5</sub></p> <p><b>Wilson, Kenneth E.; Burk, Robert M.; Biftu, Tesfaye; Ball, Richard G.; Hoogsteen, Karst;</b> Journal of Organic Chemistry; <b>vol.</b> 57; nb. 28; (1992); p. 7151 - 7158</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 28576586 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>Stage 1:</b> <b>With</b> lithium hydroxide monohydrate, water <b>in</b> tetrahydrofuran, methanol, Time= 12h, T= 20 °C</p> <p><b>Stage 2:</b> <b>With</b> water, sodium hydrogencarbonate</p> <p><b>Stage 3:</b> <b>With</b> hydrogen chloride <b>in</b> water</p> <p><b>Broeker, Joachim; Knollmueller, Max; Gaertner, Peter;</b> Tetrahedron: Asymmetry; <b>vol.</b> 20; nb. 3; (2009); p. 273 - 287</p> <p><a href="#">View in Reaxys</a></p>

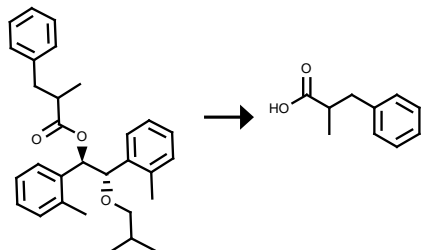

Rx-ID: 28576587 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>Stage 1:</b> <b>With</b> lithium hydroxide monohydrate, water <b>in</b> tetrahydrofuran, methanol, Time= 12h, T= 20 °C</p> <p><b>Stage 2:</b> <b>With</b> water, sodium hydrogencarbonate</p>

**Stage 3: With hydrogen chloride in water**

**Broeker, Joachim; Knollmueller, Max; Gaertner, Peter;** Tetrahedron: Asymmetry; **vol.** 20; nb. 3; (2009); p. 273 - 287

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Rx-ID: 28576624 [View in Reaxys](#)

Yield

Conditions & References

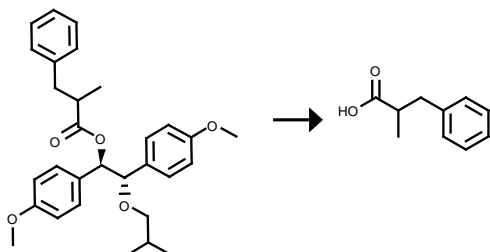
**Stage 1: With** lithium hydroxide monohydrate, water **in** tetrahydrofuran, methanol, Time= 12h, T= 20 °C

**Stage 2: With** water, sodium hydrogencarbonate

**Stage 3: With** hydrogen chloride **in** water

**Broeker, Joachim; Knollmueller, Max; Gaertner, Peter;** Tetrahedron: Asymmetry; **vol.** 20; nb. 3; (2009); p. 273 - 287

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Rx-ID: 28576625 [View in Reaxys](#)

Yield

Conditions & References

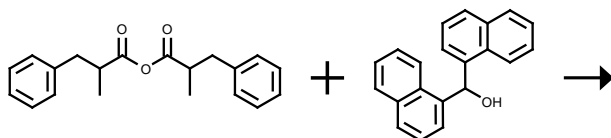
**Stage 1: With** lithium hydroxide monohydrate, water **in** tetrahydrofuran, methanol, Time= 12h, T= 20 °C

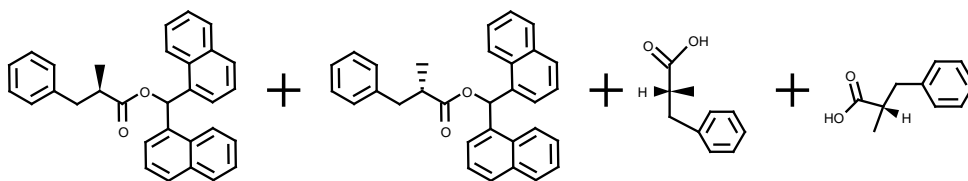
**Stage 2: With** water, sodium hydrogencarbonate

**Stage 3: With** hydrogen chloride **in** water

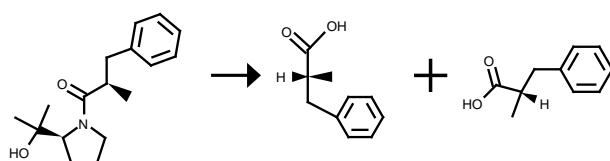
**Broeker, Joachim; Knollmueller, Max; Gaertner, Peter;** Tetrahedron: Asymmetry; **vol.** 20; nb. 3; (2009); p. 273 - 287

[View in Reaxys](#)

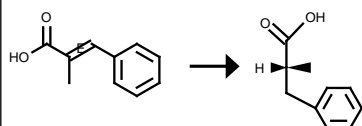



Rx-ID: 28960998 [View in Reaxys](#)

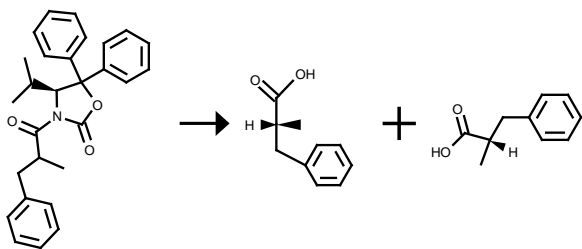
Yield	Conditions & References
	<p><b>With S-HBTM, N-ethyl-N,N-diisopropylamine in toluene, Time= 24h, T= 0 °C , optical yield given as percent ee, enantioselective reaction</b></p> <p><b>Yang, Xing; Birman, Vladimir B.;</b> Advanced Synthesis and Catalysis; <b>vol.</b> 351; nb. 14-15; (2009); p. 2301 - 2304 <a href="#">View in Reaxys</a></p>


Rx-ID: 3499086 [View in Reaxys](#)

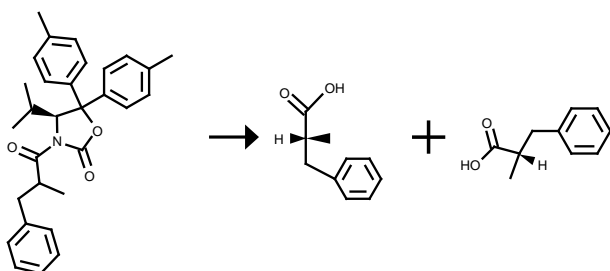
Yield	Conditions & References
	<p><b>With 3 M aq. HCl in dioxane, Time= 42h, T= 90 - 95 °C , Yield given. Yields of byproduct given. Title compound not separated from byproducts</b></p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjorn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802 <a href="#">View in Reaxys</a></p>


Rx-ID: 3962087 [View in Reaxys](#)

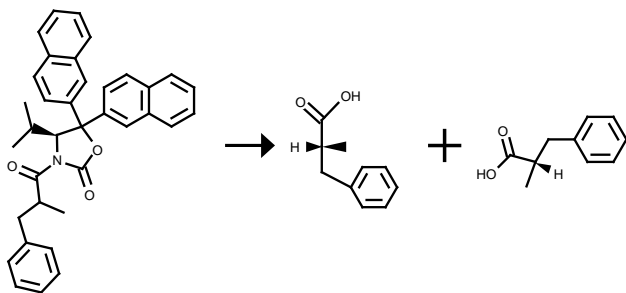
Yield	Conditions & References
61 % Spectr.	<p><b>With H<sub>2</sub>, (S,S)-&lt;(P-N)RhNBD&gt;ClO<sub>4</sub> in ethanol, Time= 3.5h, T= 30 °C , study of the asymmetric hydrogenation of various prochiral substrates with chiral ligands containing rhodium(I) complexes of ferrocenylphosphonium as asymmetric catalysts, Product distribution</b></p> <p><b>Appelton, Trevor D.; Cullen, William R.; Evans, Stephen; Kim, Tae-Jeong; Trotter, James;</b> Journal of Organometallic Chemistry; <b>vol.</b> 279; (1985); p. 5 - 22 <a href="#">View in Reaxys</a></p>
41 %Chromat.	<p><b>Example Name D20</b></p> <p><b>With hydrogen, 1-(1-dimethylaminoethyl)-2-isopropylthio-3-diphenylphosphino-ferrocene, bis(bicyclo[2.2.1]hepta-2,5-diene)rhodium(I) tetrafluoroborate in methanol, Time= 21h, T= 25 °C , p= 750.075Torr , Product distribution / selectivity</b></p> <p><b>Patent; SOLVIAS AG; WO2006/117369; (2006); (A1) German</b> <a href="#">View in Reaxys</a></p>


Rx-ID: 8848295 [View in Reaxys](#)

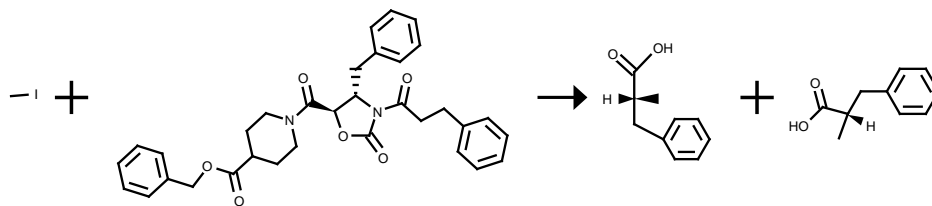
Yield	Conditions & References
	<p><b>With</b> aq. LiOH in tetrahydrofuran, Time= 25h, T= 0 - 24 °C , Title compound not separated from byproducts.</p> <p><b>Alexander, Karen; Cook, Stuart; Gibson, Colin L.; Kennedy, Alan R.;</b> Journal of the Chemical Society, Perkin Transactions 1; nb. 13; (2001); p. 1538 - 1549</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 8849963 [View in Reaxys](#)

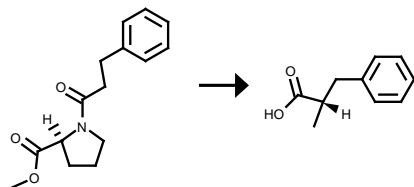
Yield	Conditions & References
	<p><b>With</b> aq. LiOH in tetrahydrofuran, Time= 25h, T= 0 - 24 °C , Title compound not separated from byproducts.</p> <p><b>Alexander, Karen; Cook, Stuart; Gibson, Colin L.; Kennedy, Alan R.;</b> Journal of the Chemical Society, Perkin Transactions 1; nb. 13; (2001); p. 1538 - 1549</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 8852653 [View in Reaxys](#)

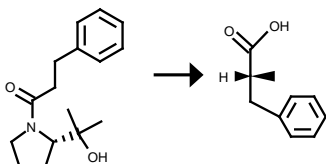
Yield	Conditions & References
	<p><b>With</b> aq. LiOH in tetrahydrofuran, Time= 25h, T= 0 - 24 °C , Title compound not separated from byproducts</p> <p><b>Alexander, Karen; Cook, Stuart; Gibson, Colin L.; Kennedy, Alan R.;</b> Journal of the Chemical Society, Perkin Transactions 1; nb. 13; (2001); p. 1538 - 1549</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 9685552 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>Stage 1:</b> With LDA in tetrahydrofuran, T= -78 - 0 °C</p> <p><b>Stage 2:</b> With LiOH, aq. H<sub>2</sub>O<sub>2</sub> in tetrahydrofuran, T= 0 °C , Further stages. Title compound not separated from by-products.</p> <p><b>Kotake, Tomoya; Rajesh, S.; Hayashi, Yoshio; Mukai, Yoshie; Ueda, Mitsuhiro; Kimura, Tooru; Kiso, Yoshia-ki;</b> Tetrahedron Letters; <b>vol.</b> 45; nb. 18; (2004); p. 3651 - 3654</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 19199538 [View in Reaxys](#)

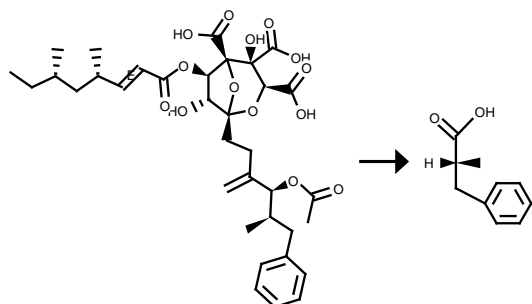
Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>2: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>3: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Mg, LDA, 3 M aq. HCl in diethyl ether, tetrahydrofuran, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjoern;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>
	<p>Reaction Steps: 3</p> <p>1: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>2: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>3: 76 percent / 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Mg, LDA, 3 M aq. HCl in diethyl ether, tetrahydrofuran, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjoern;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 19210596 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p>

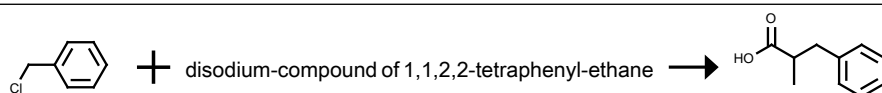
2: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C  
**With** LDA, 3 M aq. HCl in tetrahydrofuran, dioxane

**Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjørn;** Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; **vol.** 38; nb. 9; (1984); p. 795 - 802  
[View in Reaxys](#)



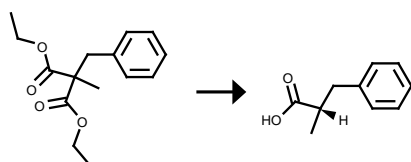
Rx-ID: 19576313 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 5</p> <p>1: 73 percent / ethyl acetate; diethyl ether / 0.08 h / -15 °C</p> <p>2: 68 percent / 0.2percent K<sub>2</sub>CO<sub>3</sub> / methanol / 1.33 h / 30 °C</p> <p>3: 89 percent / 4-(N,N-dimethylamino)pyridine / pyridine-d<sub>5</sub> / 0.05 h / 80 °C</p> <p>4: 89 percent / 3percent methanolic HCl / 16 h / Ambient temperature</p> <p><b>With</b> 0.2percent K<sub>2</sub>CO<sub>3</sub>, 4-(N,N-dimethylamino)pyridine, 3percent methanolic HCl in ethyl acetate, diethyl ether, methanol, pyridine-d<sub>5</sub></p> <p><b>Wilson, Kenneth E.; Burk, Robert M.; Biftu, Tesfaye; Ball, Richard G.; Hoogsteen, Karst;</b> Journal of Organic Chemistry; <b>vol.</b> 57; nb. 28; (1992); p. 7151 - 7158</p> <p><a href="#">View in Reaxys</a></p>



Rx-ID: 20638978 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 52 percent / NaOEt / ethanol / 5 h / Heating</p> <p>2: 74 percent / H<sub>2</sub>O/OH<sup>-</sup> / ethanol / 5 h / Heating</p> <p><b>With</b> NaOEt, H<sub>2</sub>O/OH<sup>-</sup> in ethanol</p> <p><b>Knorr, Rudolf; Lattke, Ernst;</b> Chemische Berichte; <b>vol.</b> 114; nb. 6; (1981); p. 2116 - 2131</p> <p><a href="#">View in Reaxys</a></p>



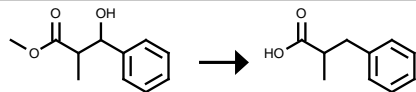
Rx-ID: 21064054 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 4</p> <p>1: aq. KOH / ethanol / 24 h / Heating</p> <p>2: dimethylformamide / 4 h / 100 - 110 °C</p> <p>3: HCl / 24 h / Ambient temperature</p>



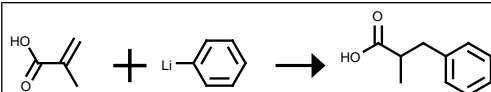
4: water / 2 h / lipase PS  
**With** aq. KOH, HCl, water **in** ethanol, dimethylformamide

**Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.;** Tetrahedron: Asymmetry; **vol.** 2; nb. 10; (1991); p. 1021 - 1030  
[View in Reaxys](#)



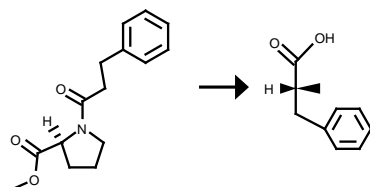
Rx-ID: 21824188 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 85 percent / Et<sub>3</sub>SiH, BF<sub>3</sub>-Et<sub>2</sub>O / CH<sub>2</sub>Cl<sub>2</sub> / 1 h / 0 °C</p> <p>2: 95 percent / aq. NaOH / methanol / 1 h / Heating</p> <p><b>With</b> Et<sub>3</sub>SiH, BF<sub>3</sub>-Et<sub>2</sub>O, aq. NaOH <b>in</b> CH<sub>2</sub>Cl<sub>2</sub>, methanol</p> <p><b>Smonou, Ioulia; Orfanopoulos, Michael;</b> Synthetic Communications; <b>vol.</b> 20; nb. 9; (1990); p. 1387 - 1397</p> <p><a href="#">View in Reaxys</a></p>



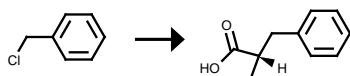
Rx-ID: 5061704 [View in Reaxys](#)

Yield	Conditions & References
12 %	<p><b>in</b> tetrahydrofuran, various solvent(s), Time= 2h, T= -70 °C , Addition</p> <p><b>Aurell, Maria Jose; Domingo, Luis Ramon; Mestres, Ramon; Munos, Elena; Zaragoza, Ramon Jose;</b> Tetrahedron; <b>vol.</b> 55; nb. 3; (1999); p. 815 - 830</p> <p><a href="#">View in Reaxys</a></p>



Rx-ID: 19199537 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 83 percent / Mg / diethyl ether / reflux, 4 h -&gt; ambient temperature</p> <p>2: LDA / tetrahydrofuran / -100 deg C, 2 h -&gt; -50 deg C</p> <p>3: 3 M aq. HCl / dioxane / 42 h / 90 - 95 °C</p> <p><b>With</b> Mg, LDA, 3 M aq. HCl <b>in</b> diethyl ether, tetrahydrofuran, dioxane</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjoern;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; <b>vol.</b> 38; nb. 9; (1984); p. 795 - 802</p> <p><a href="#">View in Reaxys</a></p>



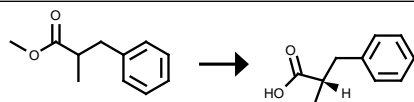
Rx-ID: 21039732 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 5</p> <p>1: 1.) NaH / 1.) ethyl ether, reflux, 2.) ethyl ether, RT, 6 h</p> <p>2: aq. KOH / ethanol / 24 h / Heating</p> <p>3: dimethylformamide / 4 h / 100 - 110 °C</p> <p>4: HCl / 24 h / Ambient temperature</p> <p>5: water / 2 h / lipase PS</p> <p><b>With</b> 1.) NaH, aq. KOH, HCl, water <b>in</b> ethanol, dimethylformamide</p> <p><b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.</b>; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030</p> <p><a href="#">View in Reaxys</a></p>



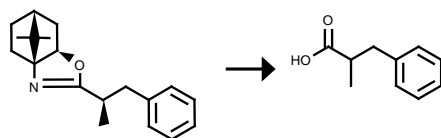
Rx-ID: 21852832 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 90 percent / Zn, I<sub>2</sub> / benzene; diethyl ether / 3 h / Heating</p> <p>2: 85 percent / Et<sub>3</sub>SiH, BF<sub>3</sub>-Et<sub>2</sub>O / CH<sub>2</sub>Cl<sub>2</sub> / 1 h / 0 °C</p> <p>3: 95 percent / aq. NaOH / methanol / 1 h / Heating</p> <p><b>With</b> Zn, I<sub>2</sub>, Et<sub>3</sub>SiH, BF<sub>3</sub>-Et<sub>2</sub>O, aq. NaOH <b>in</b> benzene, diethyl ether, CH<sub>2</sub>Cl<sub>2</sub>, methanol</p> <p><b>Smonou, Ioulia; Orfanopoulos, Michael</b>; Synthetic Communications; <b>vol.</b> 20; nb. 9; (1990); p. 1387 - 1397</p> <p><a href="#">View in Reaxys</a></p>



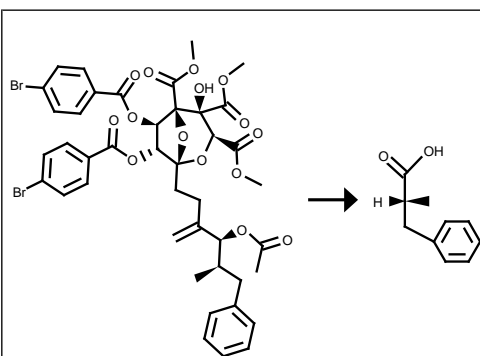
Rx-ID: 2042019 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> water, Time= 2h, lipase PS</p> <p><b>Colombo, M.; Amici, M. De; Micheli, C. De; Pitre, D.; Carrea, G.; Riva, S.</b>; Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 10; (1991); p. 1021 - 1030</p> <p><a href="#">View in Reaxys</a></p>

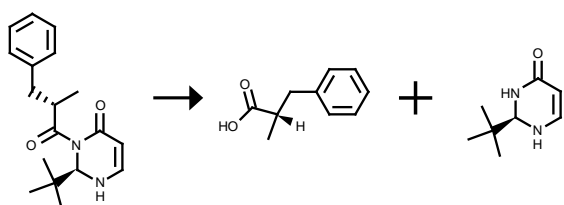


Rx-ID: 8648162 [View in Reaxys](#)

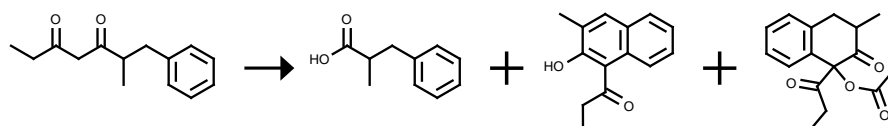
Yield	Conditions & References
	<p><b>With</b> 6 N aq. HCl, Hydrolysis</p> <p><b>Chandrasekhar, Sosale; Kausar, Amina</b>; Tetrahedron: Asymmetry; <b>vol.</b> 11; nb. 11; (2000); p. 2249 - 2254</p> <p><a href="#">View in Reaxys</a></p>


 Rx-ID: 19582334 [View in Reaxys](#)

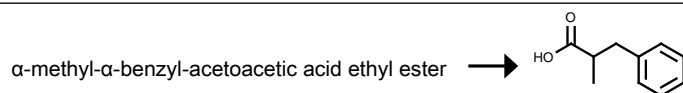
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 89 percent / 3percent methanolic HCl / 16 h / Ambient temperature</p> <p><b>With</b> 3percent methanolic HCl</p> <p><b>Wilson, Kenneth E.; Burk, Robert M.; Biftu, Tesfaye; Ball, Richard G.; Hoogsteen, Karst;</b> Journal of Organic Chemistry; <b>vol.</b> 57; nb. 28; (1992); p. 7151 - 7158</p> <p><a href="#">View in Reaxys</a></p>


 Rx-ID: 2510306 [View in Reaxys](#)

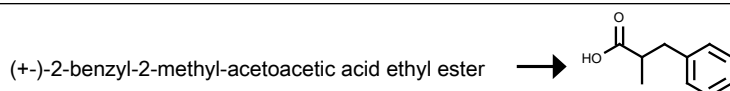
Yield	Conditions & References
	<p><b>With</b> LiOOH</p> <p><b>Negrete, George R.; Konopelski, Joseph P.;</b> Tetrahedron: Asymmetry; <b>vol.</b> 2; nb. 2; (1991); p. 105 - 108</p> <p><a href="#">View in Reaxys</a></p>


 Rx-ID: 4586308 [View in Reaxys](#)

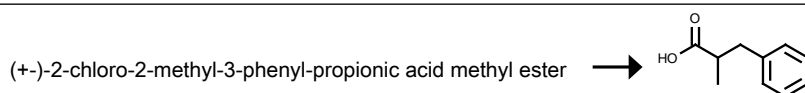
Yield	Conditions & References
9 % Chromat., 2 %, 5 %	<p><b>With</b> manganese(III) acetate, acetic acid, Time= 22h, T= 60 °C</p> <p><b>Jamie, Joanne F.; Rickards, Rodney W.;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 21; (1996); p. 2603 - 2614</p> <p><a href="#">View in Reaxys</a></p>
9 % Chromat., 5 %, 2 %	<p><b>With</b> manganese(III) acetate, acetic acid, Time= 22h, T= 60 °C</p> <p><b>Jamie, Joanne F.; Rickards, Rodney W.;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 21; (1996); p. 2603 - 2614</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 7250395 [View in Reaxys](#)

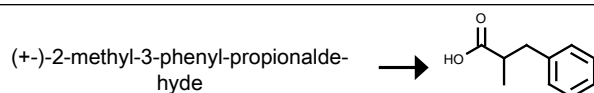
Yield	Conditions & References
	<p><b>With</b> concentrated KOH-solution</p> <p><b>Conrad; Bischoff</b>; Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 180  <a href="#">View in Reaxys</a></p>


Rx-ID: 7250396 [View in Reaxys](#)

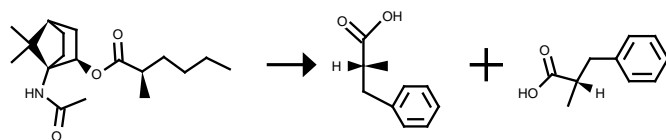
Yield	Conditions & References
	<p><b>With</b> aqueous Ba(OH)<sub>2</sub></p> <p><b>Shinya</b>; Nippon Nogei Kagaku Kaishi; <b>vol.</b> 29; (1955); p. 91,92; Chem.Abstr.; (1959); p. 1227  <a href="#">View in Reaxys</a></p>


Rx-ID: 7250397 [View in Reaxys](#)

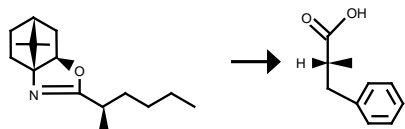
Yield	Conditions & References
	<p><b>With</b> zinc-powder, acetic acid</p> <p><b>Dombrowskii et al.</b>; Zhurnal Obshchei Khimii; <b>vol.</b> 27; (1957); p. 419; engl. Ausg. S. 473  <a href="#">View in Reaxys</a></p>


Rx-ID: 7250398 [View in Reaxys](#)

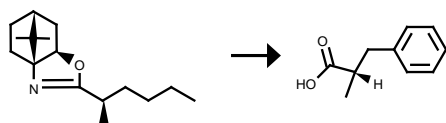
Yield	Conditions & References
	<p><b>With</b> iron isobutyrate, isobutyric acid, oxygen, T= 30 °C</p> <p><b>Patent; Shell Devel. Co.</b>; US2010358; (1933)  <a href="#">View in Reaxys</a></p>


Rx-ID: 8649941 [View in Reaxys](#)

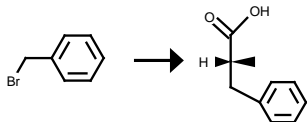
Yield	Conditions & References
	<p><b>With</b> KOH, MeOH, Heating, Hydrolysis, Title compound not separated from byproducts</p> <p><b>Chandrasekhar, Sosale; Kausar, Amina</b>; Tetrahedron: Asymmetry; <b>vol.</b> 11; nb. 11; (2000); p. 2249 - 2254  <a href="#">View in Reaxys</a></p>


Rx-ID: 15740935 [View in Reaxys](#)

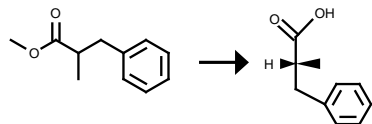
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 24 h / Heating</p> <p>2: KOH; MeOH / Heating</p> <p><b>With</b> KOH, MeOH, 1: Ring cleavage / 2: Hydrolysis</p> <p><b>Chandrasekhar, Sosale; Kausar, Amina</b>; Tetrahedron: Asymmetry; <b>vol.</b> 11; nb. 11; (2000); p. 2249 - 2254</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 15740936 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 24 h / Heating</p> <p>2: KOH; MeOH / Heating</p> <p><b>With</b> KOH, MeOH, 1: Ring cleavage / 2: Hydrolysis</p> <p><b>Chandrasekhar, Sosale; Kausar, Amina</b>; Tetrahedron: Asymmetry; <b>vol.</b> 11; nb. 11; (2000); p. 2249 - 2254</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 21560605 [View in Reaxys](#)

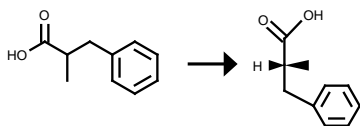
Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 1.) LDA, THF, 25 deg C, 2.) -100 deg C</p> <p>2: 92 percent / 1.0 N HCl / 2 h / 100 °C</p> <p><b>With</b> 1.0 N HCl</p> <p><b>Evans, D. A.; Takacs, J. M.</b>; Tetrahedron Letters; <b>vol.</b> 21; (1980); p. 4233 - 4236</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 21676783 [View in Reaxys](#)

Yield	Conditions & References
	<p>Reaction Steps: 2</p> <p>1: 44 percent / lipase from Pseudomonas sp. (Amano) / 3 h / Ambient temperature; 0.1 phosphate buffer, pH 7.0</p> <p>2: 92 percent / lipase from Candida cylindracea / 4.5 h / Ambient temperature; 0.1 M phosphate buffer, pH 7.0</p> <p><b>With</b> lipase from Pseudomonas sp. (Amano), lipase from Candida cylindracea</p>

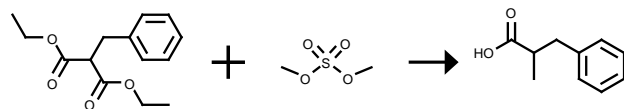
**Delnick, Deborah L.; Margolin, Alexey L.;** Tetrahedron Letters; **vol.** 31; nb. 47; (1990); p. 6797 - 6798

[View in Reaxys](#)



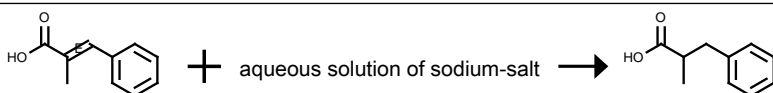
Rx-ID: 438147 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> quinine</p> <p><b>Kipping; Hunter;</b> Journal of the Chemical Society; <b>vol.</b> 83; (1903); p. 1007</p> <p><a href="#">View in Reaxys</a></p>



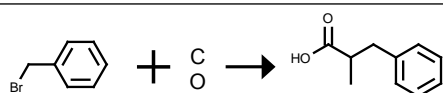
Rx-ID: 4271420 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> 1.) NaH, 2.) KOH, 1.) THF, 3 h, room temp., 2.) H<sub>2</sub>O, room temp., 3.) 190 deg C, Yield given. Multistep reaction</p> <p><b>Bucher, von Christoph B.; Linden, Anthony; Heimgartner, Heinz;</b> Helvetica Chimica Acta; <b>vol.</b> 78; nb. 4; (1995); p. 935 - 946</p> <p><a href="#">View in Reaxys</a></p>



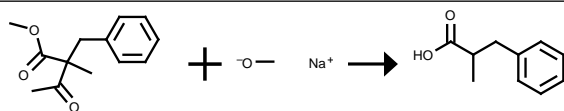
Rx-ID: 7250400 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> sodium-amalgam, Hydrogenation</p> <p><b>Holden; Lapworth;</b> Journal of the Chemical Society; (1931); p. 2368,2375</p> <p><a href="#">View in Reaxys</a></p>

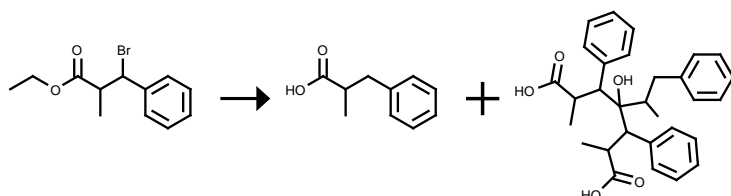


Rx-ID: 17758483 [View in Reaxys](#)

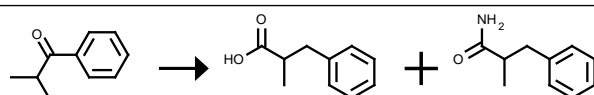
Yield	Conditions & References
	<p>Reaction Steps: 3</p> <p>1: 1.) NaOEt / 1.) EtOH, 30 min, 2.) EtOH, reflux, 2 h</p> <p>2: aq. KOH / 1 h / Heating</p> <p>3: conc. H<sub>2</sub>SO<sub>4</sub> / 1 h / Heating</p> <p><b>With</b> 1.) NaOEt, aq. KOH, conc. H<sub>2</sub>SO<sub>4</sub></p> <p><b>Tyrrell, Elizabeth; Tsang, Michael W. H.; Skinner, George A.; Fawcett, John;</b> Tetrahedron; <b>vol.</b> 52; nb. 29; (1996); p. 9841 - 9852</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 470487 [View in Reaxys](#)

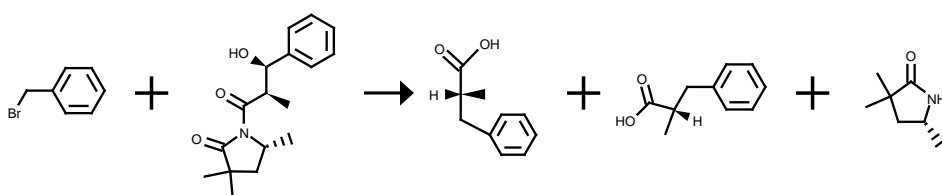
Yield	Conditions & References
	<p><b>With</b> methanol, Behandeln des Reaktionsprodukts mit wss. Natronlauge</p> <p><b>Patent:</b> Kay-Fries Chem. Inc.; US2413493; (1941)</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 472551 [View in Reaxys](#)

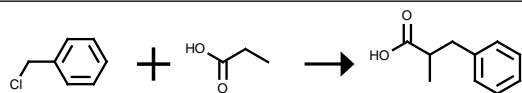
Yield	Conditions & References
	<p><b>With</b> magnesium, nachfolgend Verseifen</p> <p><b>Salkind; Grabowski;</b> Zhurnal Russkago Fiziko-Khimicheskago Obshchestva; <b>vol.</b> 46; (1914); p. 505; Chem. Zentralbl.; <b>vol.</b> 85; nb. II; (1914); p. 1270</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 796180 [View in Reaxys](#)

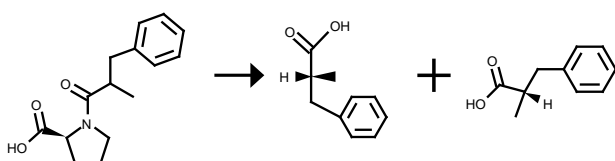
Yield	Conditions & References
	<p><b>With</b> yellow aqueous ammonium sulfide, T= 200 °C , unter Druck</p> <p><b>Willgerodt; Merk;</b> Journal fuer Praktische Chemie (Leipzig); <b>vol.</b> &lt;2&gt; 80; (1909); p. 193</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2400803 [View in Reaxys](#)

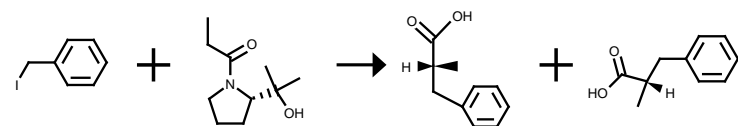
Yield	Conditions & References
	<p><b>With</b> 1.) LDA, 2.) LiOH, 1.) THF, 0 deg C, Yield given. Multistep reaction. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Davies, Stephen G.; Doisneau, Gilles J.-M.; Prodger, Jeremy C.; Sangane, Hitesh J.;</b> Tetrahedron Letters; <b>vol.</b> 35; nb. 15; (1994); p. 2373 - 2376</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2696287 [View in Reaxys](#)

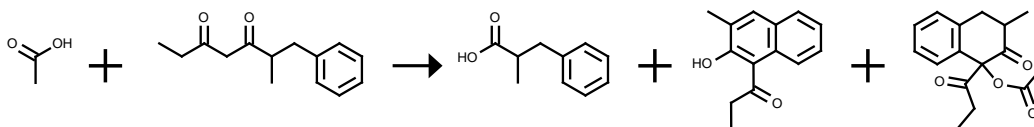
Yield	Conditions & References
	<p><b>With</b> 1) BuLi, diisopropylamine, 1) THF, -60 deg C to room temp., 30 min; 2) THF, -40 deg C to 20 deg C, Yield given. Multistep reaction</p> <p><b>Boche, Gernot; Buckl, Klaus; Martens, Diether; Schneider, Dieter R.;</b> Liebigs Annalen der Chemie; nb. 7; (1980); p. 1135 - 1171  <a href="#">View in Reaxys</a></p>


Rx-ID: 9618144 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> glacial acetic acid, HCl, Heating, Title compound not separated from byproducts</p> <p><b>Srivastava, Stuti; Goswami, Lalit N.; Dikshit, Dinesh K.;</b> Indian Journal of Chemistry, Section B: Organic Chemistry Including Medicinal Chemistry; vol. 42; nb. 10; (2003); p. 2628 - 2631  <a href="#">View in Reaxys</a></p>

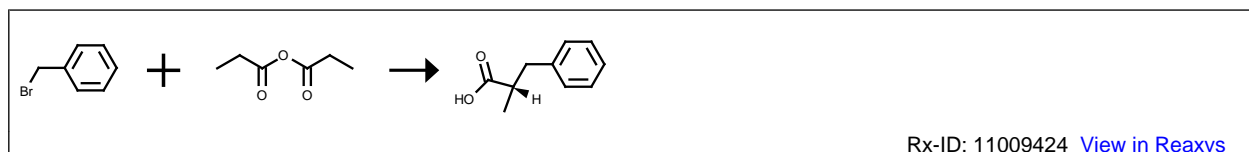
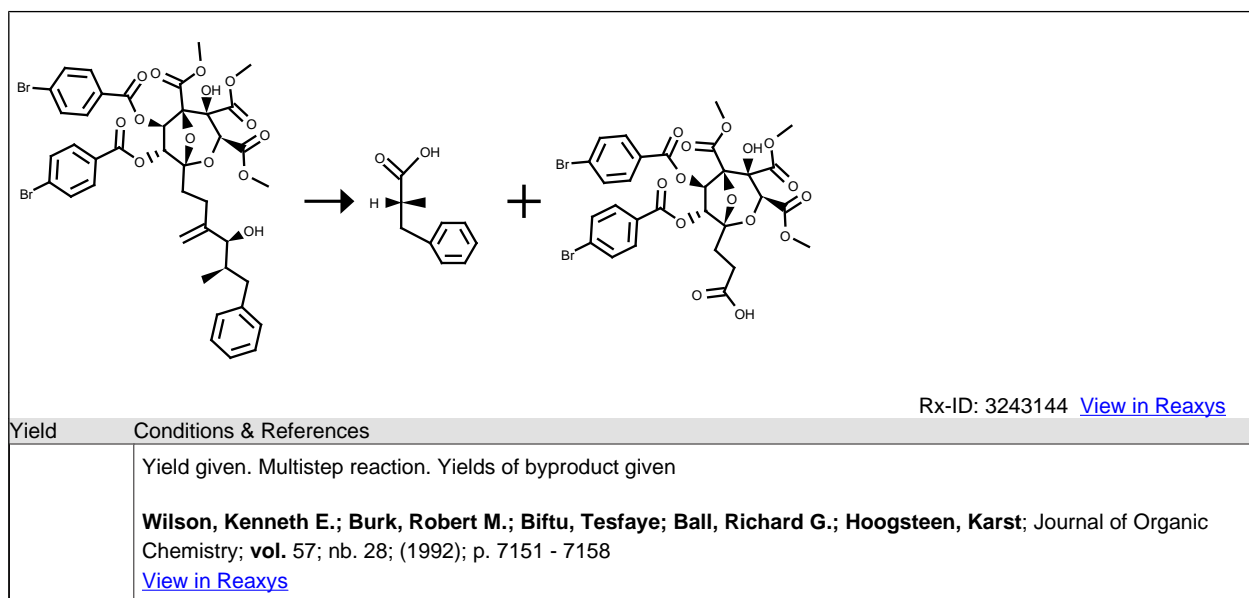
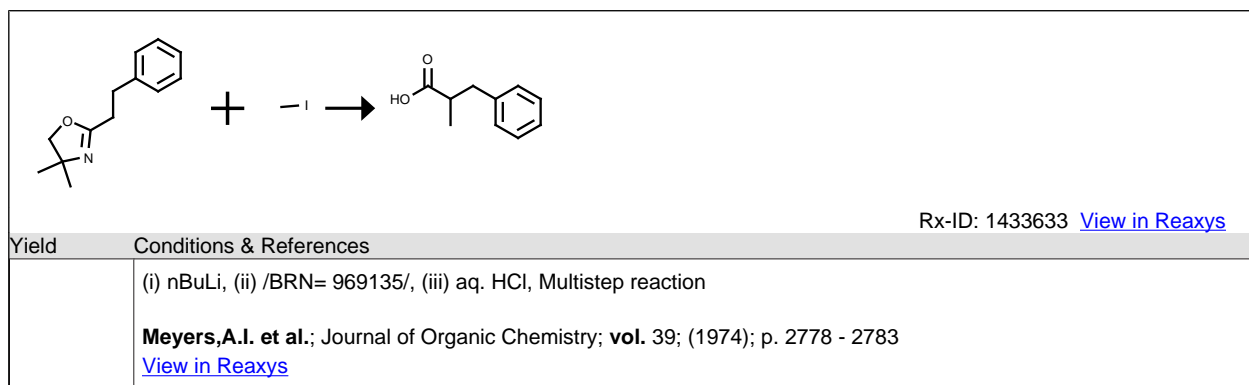
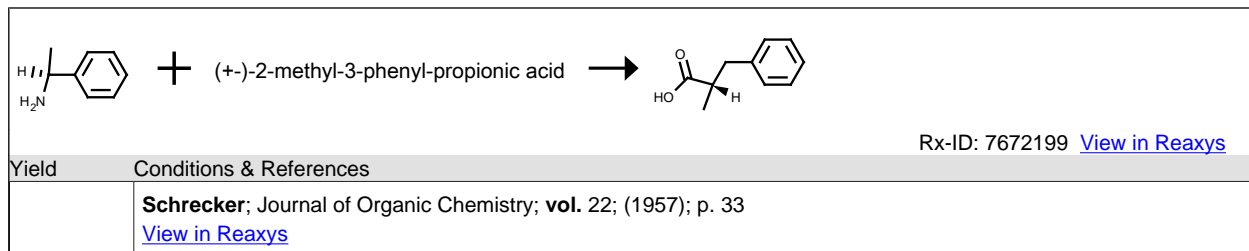
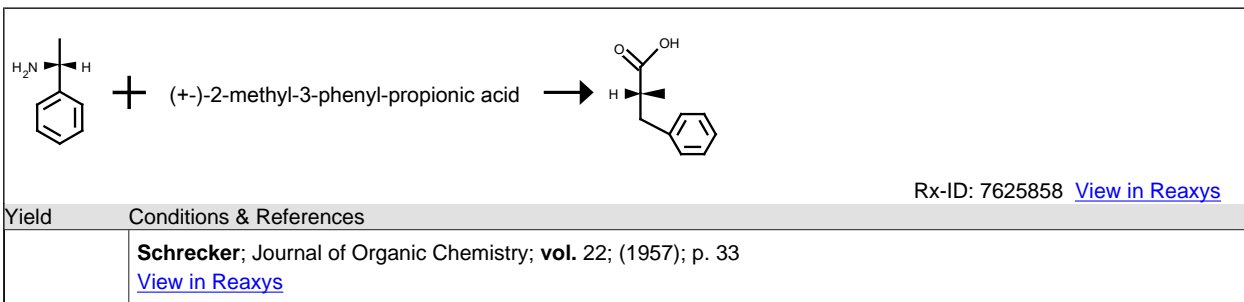

Rx-ID: 3385499 [View in Reaxys](#)

Yield	Conditions & References
	<p><b>With</b> 1.) LDA, 2.) 3 M aq. HCl, 1.) THF, 0 deg C, 2 h, 2.) dioxane, 90-95 deg C, 42 h, Yield given. Multistep reaction. Yields of byproduct given. Title compound not separated from byproducts</p> <p><b>Guoqiang, Lin; Hjalmarsson, Mats; Hoegberg, Hans-Erik; Jernstedt, Karen; Norin, Torbjorn;</b> Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; vol. 38; nb. 9; (1984); p. 795 - 802  <a href="#">View in Reaxys</a></p>

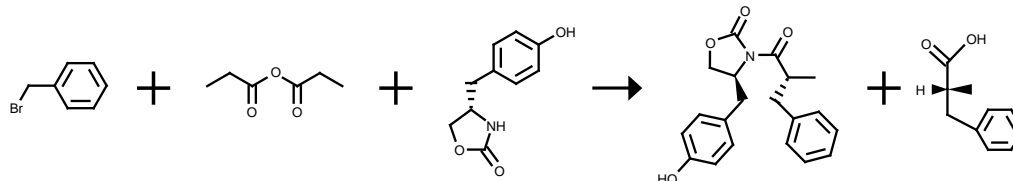

Rx-ID: 4575471 [View in Reaxys](#)

Yield	Conditions & References
2 %, 5 %, 9 % Chromat.	<p><b>With</b> manganese(III) acetate, Time= 22h, T= 60 °C</p> <p><b>Jamie, Joanne F.; Rickards, Rodney W.;</b> Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999); nb. 21; (1996); p. 2603 - 2614  <a href="#">View in Reaxys</a></p>

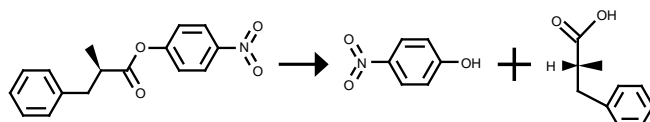




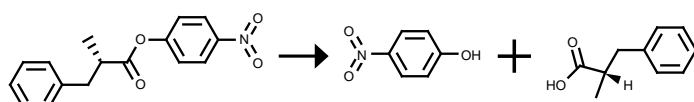
Yield	Conditions & References
90 mg	<p>Multistep reaction.</p> <p><b>McGhee, Andrea M.; Kizirian, Jean-Claude; Procter, David J.</b>; Organic and Biomolecular Chemistry; <b>vol.</b> 5; nb. 7; (2007); p. 1021 - 1024  <a href="#">View in Reaxys</a></p>


Rx-ID: 26018125 [View in Reaxys](#)

Yield	Conditions & References
	<p>Multistep reaction.</p> <p><b>Green, Rachel; Merritt, Andrew T.; Bull, Steven D.</b>; Chemical Communications (Cambridge, United Kingdom); nb. 4; (2008); p. 508 - 510  <a href="#">View in Reaxys</a></p>


Rx-ID: 3108371 [View in Reaxys](#)

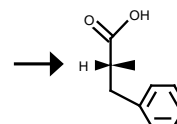
Yield	Conditions & References
	<p><b>With</b> H<sub>2</sub>O, phosphate buffer <b>in</b> acetonitrile, T= 25 °C , α-chymotrypsin as catalyst, pH: 7.5; stereospecificity in hydrolysis of acylated α-chymotrypsins, Rate constant, Mechanism</p> <p><b>Oetvoes, Laszlo; Kraicsovits, Ferenc</b>; Tetrahedron; <b>vol.</b> 48; nb. 23; (1992); p. 5009 - 5014  <a href="#">View in Reaxys</a></p>


Rx-ID: 3108372 [View in Reaxys](#)

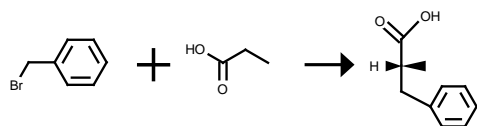
Yield	Conditions & References
	<p><b>With</b> H<sub>2</sub>O, phosphate buffer <b>in</b> acetonitrile, T= 25 °C , α-chymotrypsin as catalyst, pH: 7.5; stereospecificity in hydrolysis of acylated α-chymotrypsins, Rate constant, Mechanism</p> <p><b>Oetvoes, Laszlo; Kraicsovits, Ferenc</b>; Tetrahedron; <b>vol.</b> 48; nb. 23; (1992); p. 5009 - 5014  <a href="#">View in Reaxys</a></p>



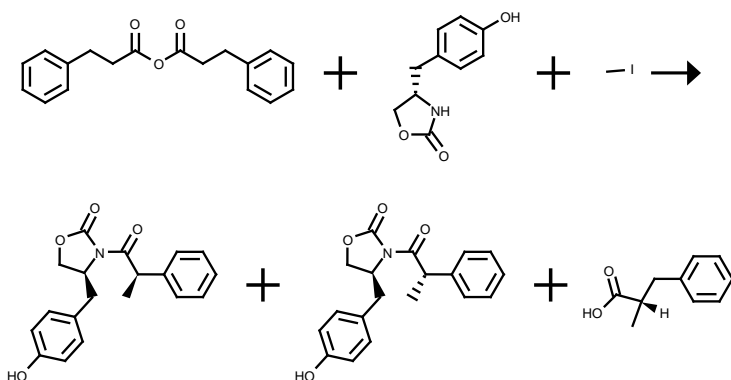
aq.-ethanolic NaOH-solution


Rx-ID: 7625859 [View in Reaxys](#)

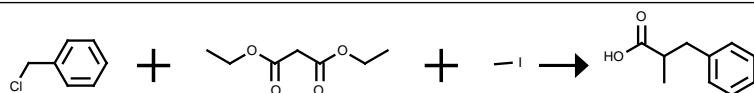
Yield	Conditions & References
	<p><b>Kenyon; Phillips; Pittman</b>; Journal of the Chemical Society; (1935); p. 1079  <a href="#">View in Reaxys</a></p>


Rx-ID: 9883488 [View in Reaxys](#)

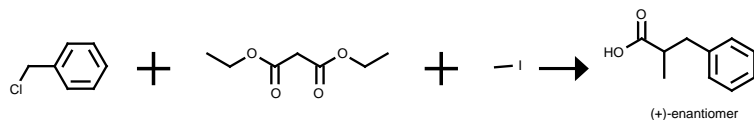
Yield	Conditions & References
16.6 mg	<p>Multistep reaction</p> <p><b>Kotake, Tomoya; Hayashi, Yoshio; Rajesh, S.; Mukai, Yoshie; Takiguchi, Yuka; Kimura, Tooru; Kiso, Yoshiaki;</b> Tetrahedron; <b>vol.</b> 61; nb. 15; (2005); p. 3819 - 3834</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 26018123 [View in Reaxys](#)

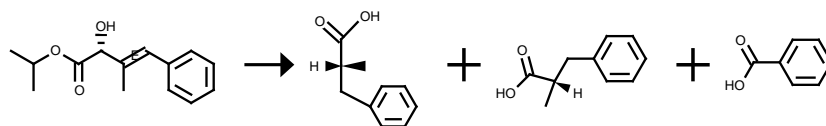
Yield	Conditions & References
	<p>Multistep reaction.</p> <p><b>Green, Rachel; Merritt, Andrew T.; Bull, Steven D.;</b> Chemical Communications (Cambridge, United Kingdom); nb. 4; (2008); p. 508 - 510</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 1203266 [View in Reaxys](#)

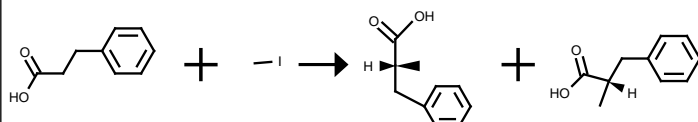
Yield	Conditions & References
	<p><b>Kashiwagi, T. et al.;</b> Tetrahedron; <b>vol.</b> 26; (1970); p. 3619 - 3629</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 1203267 [View in Reaxys](#)

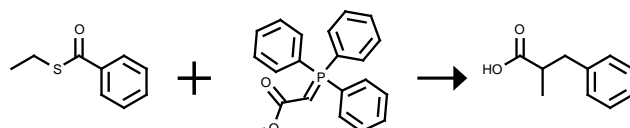
Yield	Conditions & References
	<p><b>Kashiwagi, T. et al.;</b> Tetrahedron; <b>vol.</b> 26; (1970); p. 3619 - 3629</p> <p><a href="#">View in Reaxys</a></p>


Rx-ID: 2792917 [View in Reaxys](#)

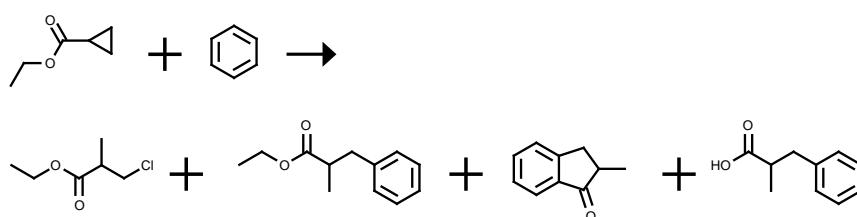
Yield	Conditions & References
	Yield given. Multistep reaction. Yields of byproduct given
	<b>Yu, Hongtao; Simon, Helmut</b> ; Tetrahedron; <b>vol.</b> 47; nb. 43; (1991); p. 9035 - 9052 <a href="#">View in Reaxys</a>


Rx-ID: 9901344 [View in Reaxys](#)

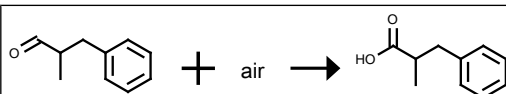
Yield	Conditions & References
	Multistep reaction. Title compound not separated from byproducts
	<b>Kotake, Tomoya; Hayashi, Yoshio; Rajesh, S.; Mukai, Yoshie; Takiguchi, Yuka; Kimura, Tooru; Kiso, Yoshiaki</b> ; Tetrahedron; <b>vol.</b> 61; nb. 15; (2005); p. 3819 - 3834 <a href="#">View in Reaxys</a>


Rx-ID: 1340292 [View in Reaxys](#)

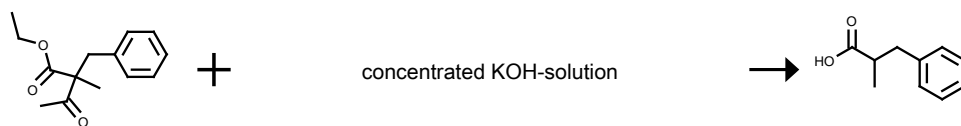
Yield	Conditions & References
	(i) Raney-Ni, THF, (ii) KOH, aq. EtOH, (iii) H <sub>2</sub> , Raney-Ni, aq. MeOH, Multistep reaction
	<b>Bestmann, H.J. et al.</b> ; Chemische Berichte; <b>vol.</b> 99; (1966); p. 1906 - 1911 <a href="#">View in Reaxys</a>


Rx-ID: 3917528 [View in Reaxys](#)

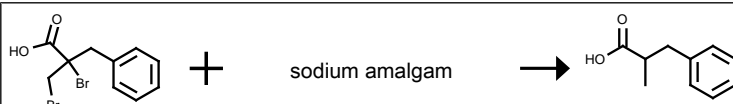
Yield	Conditions & References
	With AlCl <sub>3</sub> , Ambient temperature, Mechanism, Product distribution
	<b>Pinnick, Harold W.; Brown, Stephen P.; McLean, Elizabeth A.; Zoller, Linwood W.</b> ; Journal of Organic Chemistry; <b>vol.</b> 46; nb. 18; (1981); p. 3758 - 3760 <a href="#">View in Reaxys</a>


Rx-ID: 7250399 [View in Reaxys](#)

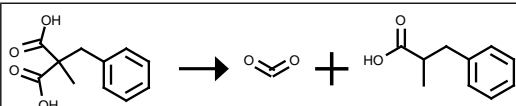
Yield	Conditions & References
	<b>v. Miller; Rohde</b> ; Chemische Berichte; <b>vol.</b> 23; (1890); p. 1080 <a href="#">View in Reaxys</a>


 Rx-ID: 7987451 [View in Reaxys](#)

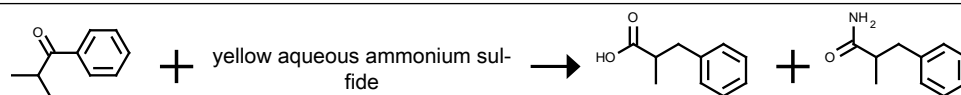
Yield	Conditions & References
	<b>Bischoff; Conrad</b> ; Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 179,180 <a href="#">View in Reaxys</a>


 Rx-ID: 7987452 [View in Reaxys](#)

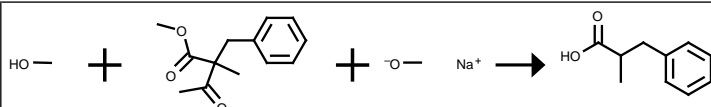
Yield	Conditions & References
	<b>Simonsen</b> ; Journal of the Chemical Society; <b>vol.</b> 117; (1920); p. 569 <a href="#">View in Reaxys</a>


 Rx-ID: 5809826 [View in Reaxys](#)

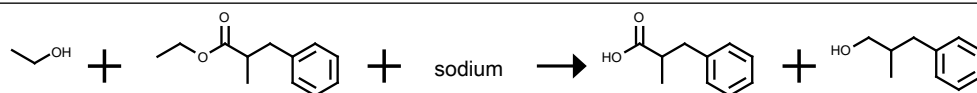
Yield	Conditions & References
	<b>Conrad; Bischoff</b> ; Justus Liebigs Annalen der Chemie; <b>vol.</b> 204; (1880); p. 178 <a href="#">View in Reaxys</a>


 Rx-ID: 5815025 [View in Reaxys](#)

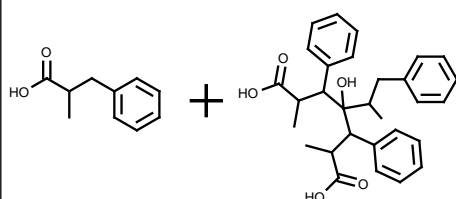
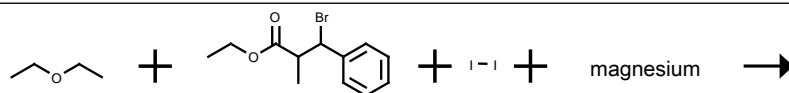
Yield	Conditions & References
	T= 200 °C , unter Druck <b>Willgerodt; Merk</b> ; Journal fuer Praktische Chemie (Leipzig); <b>vol.</b> <2> 80; (1909); p. 193 <a href="#">View in Reaxys</a>


 Rx-ID: 55936 [View in Reaxys](#)

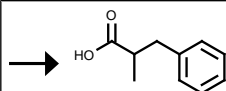
Yield	Conditions & References
	Behandeln des danach isolierten Reaktionsprodukts mit wss. Natronlauge <b>Patent; Kay-Fries Chem. Inc.</b> ; US2413493; (1941) <a href="#">View in Reaxys</a>


Rx-ID: 7047562 [View in Reaxys](#)

Yield	Conditions & References
	<b>v. Braun; Grabowski; Kirschbaum</b> ; Chemische Berichte; <b>vol.</b> 46; (1913); p. 1280 <a href="#">View in Reaxys</a>

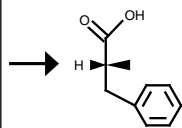

Rx-ID: 8202788 [View in Reaxys](#)

Yield	Conditions & References
	nachfolgend Verseifen des Reaktionsproduktes  <b>Salkind; Grabowski</b> ; Zhurnal Russkago Fiziko-Khimicheskago Obshchestva; <b>vol.</b> 46; (1914); p. 505; Chem. Zentralbl.; <b>vol.</b> 85; nb. II; (1914); p. 1270 <a href="#">View in Reaxys</a>


Rx-ID: 7250394 [View in Reaxys](#)

Yield	Conditions & References
99%	Example Name 1 Example Title EXAMPLE 1 The benzylmethylmalonic acid (49.2 g) was dissolved in 400 ml of acetonitrile with 1.69 g of cuprous oxide and heated to reflux for 5 hours. The solvent was removed under vacuum. The residue was taken up in 400 ml of diethyl ether and rinsed with 10percent hydrochloric acid (*300 ml), 300 ml of saturated sodium chloride, dried over magnesium sulfate, filtered and concentrated. The residue was purified by flash chromatography (5percentto 10percent methanol in chloroform) to yield 38.3 g of 2-benzylpropionic acid (99percent yield).  <b>Patent; Merrell Pharmaceuticals Inc.</b> ; US5840729; (1998); (A1) English <a href="#">View in Reaxys</a>
99%	Example Name 1 Example Title EXAMPLE 1 The benzylmethylmalonic acid (49.2 g) was dissolved in 400 ml of acetonitrile with 1.69 g of cuprous oxide and heated to reflux for 5 hours. The solvent was removed under vacuum. The residue was taken up in 400 ml of diethyl ether and rinsed with 10percent hydrochloric acid (2*300 ml), 300 ml of saturated sodium chloride, dried over magnesium sulfate, filtered and concentrated. The residue was purified by flash chromatography (5percentando10percent methanol in chloroform) to yield 38.3 g of 2-benzylpropionic acid (99percent yield).  <b>Patent; Merrell Dow Pharmaceuticals Inc.</b> ; US5047534; (1991); (A1) English <a href="#">View in Reaxys</a>

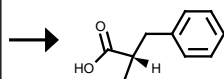
99%	<p>Example Name 8 Example Title EXAMPLE 8</p> <p>The benzylmethylmalonic acid (49.2 g) was dissolved in 400 ml of acetonitrile, treated with 1.69 g of cuprous oxide and heated to reflux for 5 hours.</p> <p>The solvent was removed under vacuum and the residue taken up in 400 ml of diethyl ether and rinsed with 10percent hydrochloric acid (2*300 ml), saturated sodium chloride (300 ml), dried over magnesium sulfate, filtered and concentrated.</p> <p>The residue was purified by flash chromatography (5percent to 10percent methanol in chloroform) to yield 38.37 g of 2-benzylpropionic acid (99percent yield).</p> <p><b>Patent; Merrell Dow Pharmaceuticals Inc.;</b> US5047534; (1991); (A1) English <a href="#">View in Reaxys</a></p> <p><b>Patent; Merrell Pharmaceuticals Inc.;</b> US5840729; (1998); (A1) English <a href="#">View in Reaxys</a></p>
	<p><b>Reinheckel,H.; Tauber,G.;</b> Monatshefte fuer Chemie; <b>vol.</b> 98; (1967); p. 1944 - 1953 <a href="#">View in Reaxys</a></p> <p><b>Patent; Farbwerke Hoechst A.G.;</b> DE1300565; (1965); Chem.Abstr.; <b>vol.</b> 71; nb. 91098t; (1969) <a href="#">View in Reaxys</a></p> <p><b>Harmon et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 34; (1969); p. 3684 <a href="#">View in Reaxys</a></p> <p><b>Normant; Angelo;</b> Bulletin de la Societe Chimique de France; (1962); p. 814 <a href="#">View in Reaxys</a></p> <p><b>Hjelte;</b> Acta Chemica Scandinavica (1947-1973); <b>vol.</b> 15; (1961); p. 1200 <a href="#">View in Reaxys</a></p> <p><b>Spassov; Stefanova;</b> Journal of Molecular Structure; <b>vol.</b> 53; (1979); p. 219,220,223 <a href="#">View in Reaxys</a></p> <p><b>Larcheveque;</b> Annales de Chimie (Cachan, France); <b>vol.</b> 5; nb. 14; (1970); p. 129,132 <a href="#">View in Reaxys</a></p> <p><b>Kuchar et al.;</b> Collection of Czechoslovak Chemical Communications; <b>vol.</b> 44; (1979); p. 183,186 <a href="#">View in Reaxys</a></p> <p><b>Aviron-Violet et al.;</b> Journal of Molecular Catalysis; <b>vol.</b> 5; (1979); p. 44; Chem.Abstr.; <b>vol.</b> 91; nb. 57462k <a href="#">View in Reaxys</a></p> <p><b>Aguiar et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 41; nb. 9; (1976); p. 1545,1547 <a href="#">View in Reaxys</a></p> <p><b>Byers; Wolfenden;</b> Biochemistry; <b>vol.</b> 12; (1973); p. 2070 <a href="#">View in Reaxys</a></p> <p><b>Marvell et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 25; (1960); p. 608,610 <a href="#">View in Reaxys</a></p> <p><b>Marshall et al.;</b> Journal of Organic Chemistry; <b>vol.</b> 31; (1966); p. 4315,4317 <a href="#">View in Reaxys</a></p> <p><b>Horner et al.;</b> Liebigs Annalen der Chemie; (1979); p. 341,351 <a href="#">View in Reaxys</a></p> <p><b>Pastuschak; Dombrowskii;</b> J. Gen. Chem. USSR (Engl. Transl.); <b>vol.</b> 34; (1964); p. 3150; Zhurnal Obshchei Khimii; <b>vol.</b> 34; (1964); p. 3110 <a href="#">View in Reaxys</a></p> <p><b>Gossauer; Ossorio;</b> Anales de la Real Sociedad Espanola de Fisica y Quimica, Serie B: Quimica; <b>vol.</b> 59; (1963); p. 185,189 <a href="#">View in Reaxys</a></p> <p><b>Larcheveque;</b> Comptes Rendus des Seances de l'Academie des Sciences, Serie C: Sciences Chimiques; <b>vol.</b> 268; (1969); p. 640 <a href="#">View in Reaxys</a></p>
	<p>Example Name 8 Example Title EXAMPLE 8</p> <p>After six additional recrystallizations from 50percent aqueous ethanol there remained 18.8 g of the quinine salt.</p> <p>The mother liquors from the above recrystallizations were acidified and extracted to yield 24.86 g of recovered 2-benzylpropionic acid.</p> <p><b>Patent; Merrell Dow Pharmaceuticals Inc.;</b> US5047534; (1991); (A1) English <a href="#">View in Reaxys</a></p> <p><b>Patent; Merrell Pharmaceuticals Inc.;</b> US5840729; (1998); (A1) English</p>


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Yield	Conditions & References
91%	<p>Example Name 19 Example Title EXAMPLE 19 STR45 EXAMPLE 19 STR45</p> <p>A 10 mL recovery flask equipped with a magnetic stirrer was charged with [1S-[1R*(S*), 2R*]]-N-(2-hydroxy-1-methyl-2-phenylethyl)-N,2-dimethyl benzenepropion amide (75 mg, 0.24 mmol, 1.0 equiv) and a mixture of tetrabutylammonium hydroxide (0.78g of a 40percent aqueous solution, 1.21 mmol, 5 equiv) in 4:1 water: tert-butanol (5 mL), and the resulting mixture was refluxed for 22 h under a Liebig condenser.</p> <p>The reaction mixture was basified with 1N sodium hydroxide (100 mL), and extracted with ether (3*10 mL).</p> <p>The aqueous phase was acidified with 3N HCl, saturated with sodium chloride, and extracted with ethyl acetate (3*15 mL).</p> <p>The combined ethyl acetate extracts were washed once with water (5 mL), dried over sodium sulfate, filtered, and concentrated to give R-α-methyl benzenepropionic acid (36 mg, 91percent yield) with an enantiomeric excess of 94percent (as determined by chiral GC analysis of the R-α-methylbenzyl amide of the acid).</p> <p><sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ7.25 (m, 5H), 3.09 (dd, 1H, J<sub>1</sub>=6.1 Hz, J<sub>2</sub>=13.1 Hz), 2.75 (m, 2H, H), 1.18 (d, 3H, J=6.8 Hz); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) δ182.5, 139.0, 129.0, 128.4, 126.4, 41.2, 39.3, 16.5; FTIR (neat film) cm<sup>-1</sup> 2976 (s,br, OH), 2657 (m, br), 1948 (w), 1877 (w), 1806 (w), 1707 (s, C=O), 1496 (m), 1454 (s), 1417 (m), 1294 (s), 1241 (s), 1117 (w), 1082 (w), 942 (m), 744 (m), 700 (s), 549 (w).</p> <p><b>Patent:</b> California Institute of Technology; US5488131; (1996); (A1) English <a href="#">View in Reaxys</a></p>
87%	<p>Example Name 18 Example Title R-α-methyl benzenepropionic acid EXAMPLE 18 STR44</p> <p>R-α-methyl benzenepropionic acid</p> <p>A 50 mL round-bottomed flask was charged with [1S-[1R*(S*), 2R*]]-N-(2-hydroxy-1-methyl-2-phenylethyl)-N,2-dimethyl benzenepropion amide (1.2458g, 4.0 mmol) and 10:8 18N sulfuric acid: dioxane (18 mL).</p> <p>The resulting mixture was refluxed for 1 h under a Liebig condenser.</p> <p>The mixture was then basified with 50percent aqueous sodium hydroxide, and washed with dichloromethane (2*50 mL).</p> <p>The remaining aqueous phase was acidified with 6N aqueous sulfuric acid and extracted with dichloromethane (3*50 mL).</p> <p>These extracts were dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure to afford R-α-methyl benzenepropionic acid (0.5719 g, 87percent yield) with an enantiomeric excess of 97percent (as determined by chiral GC analysis of the R-α-methylbenzyl amide of the acid).</p> <p><b>Patent:</b> California Institute of Technology; US5488131; (1996); (A1) English <a href="#">View in Reaxys</a></p>
81%	<p>Example Name 8 Example Title Preparation of Some Representative Intermediates for the Elaboration of the N-Terminus of the Compound of Formula 1</p> <p>To a cooled solution (0.deg.) of the latter oxazolidine derivative (26.7 g, 97.0 mmol) in THF (9500 mL) and H<sub>2</sub> O (1.5 L) was added a 30percent aqueous solution of hydrogen peroxide (55 mL, 0.5 mol), followed by the addition of a solution of LiOH.H<sub>2</sub> O (8.67 g, 200 mmol) in H<sub>2</sub> O (15 mL).</p> <p>The resultant mixture was vigorously stirred for 1 h at 0.deg..</p> <p>A solution of Na<sub>2</sub> SO<sub>3</sub> (100 g) in H<sub>2</sub> O (700 mL) and solid NaHCO<sub>3</sub> (20 g) were added sequentially.</p> <p>After 5 min, the THF was removed under reduce pressure.</p> <p>The residual aqueous solution was washed with CH<sub>2</sub> Cl<sub>2</sub> (3 X).</p> <p>The aqueous phase was rendered acidic with 10percent aqueous HCl<sub>1</sub> and extracted with Et<sub>2</sub> O (3 X).</p> <p>The combined Et<sub>2</sub> O extracts were washed with brine, dried (MgSO<sub>4</sub>) and concentrated to afford α(R)-methylbenzenepropionic acid as a clear liquid (12.8 g, 81percent yield).</p>



	<sup>1</sup> H NMR (CDCl <sub>3</sub> ) δ 7.33-7.19 (m, 5H), 3.10 (dd, J=13.5, 6.5, 1H), 2.83-2.74 (m, 1H), 2.79 (dd, J=13.5, 8 Hz, 1H), 1.20 (d, J=7 Hz, 3H).  <b>Patent; Bio-Mega/Boehringer Ingelheim Research, Inc.;</b> US5672586; (1997); (A1) English <a href="#">View in Reaxys</a>
	<b>Meyers, A.I. et al.;</b> Heterocycles; <b>vol.</b> 6; (1977); p. 971 - 977 <a href="#">View in Reaxys</a> <b>Valentine; Scott;</b> Synthesis; (1978); p. 329,343 <a href="#">View in Reaxys</a> <b>Bezmer; Brown;</b> Journal of Pharmaceutical Sciences; <b>vol.</b> 60; (1971); p. 583,584 <a href="#">View in Reaxys</a>
	<p>Example Name 8  Example Title EXAMPLE 8  This acid was combined with 18.4 g of d-(+)-α-methylbenzylamine in 160 ml of ethyl acetate, dissolved by heating on a steam bath, cooled, and the precipitate was collected to yield 35 g of the amine salt.  After three additional recrystallizations from ethyl acetate, the amine salt (0.4 g) was treated with 100 ml of 1M sulfuric acid.  The aqueous layer was extracted with chloroform (2*100 ml) and the combined organic extracts dried over magnesium sulfate, filtered and concentrated.  The residue was purified by radial chromatography (5percent methanol in chloroform, 2 mm plate) to yield 186 mg of (R)-2-benzylpropionic acid.</p> <p><b>Patent; Merrell Pharmaceuticals Inc.;</b> US5840729; (1998); (A1) English  <a href="#">View in Reaxys</a></p>
	<p>Example Name 8  Example Title EXAMPLE 8  This acid was combined with 18.4 g of d-(+)-α-methylbenzylamine in 160 ml of ethyl acetate, dissolved by heating on a steam bath, cooled, and the precipitate was collected to yield 35 g of the amine salt.  After three additional recrystallizations from ethyl acetate, the amine salt (0.4 g) was treated with 100 ml of 1M sulfuric acid.  The aqueous layer was extracted with chloroform (2 x 100 ml) and the combined organic extracts dried over magnesium sulfate, filtered and concentrated.  The residue was purified by radial chromatography (5percent methanol in chloroform, 2 mm plate) to yield 186 mg of (R)-2-benzylpropionic acid.</p> <p><b>Patent; Merrell Dow Pharmaceuticals Inc.;</b> US5047534; (1991); (A1) English  <a href="#">View in Reaxys</a></p>


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Yield	Conditions & References
	<b>Valentine; Scott;</b> Synthesis; (1978); p. 329,343 <a href="#">View in Reaxys</a>
	<p>Example Name 1  Example Title EXAMPLE 1  After six additional recrystallizations from 50percent aqueous ethanol, there remained 18.8 g of the quinine salt.  The quinine salt (0.34 g) was treated with 100 ml of 1M sulfuric acid and extracted with chloroform (2*100 ml).  The combined organic extracts were dried over magnesium sulfate, filtered and concentrated.  The residue was purified by radial chromatography (5percent to 10percent methanol in chloroform, 2 mm plate) to yield 89 mg of (S)-2-methyl-3-phenylpropionic acid.</p> <p><b>Patent; Merrell Dow Pharmaceuticals Inc.;</b> US5047534; (1991); (A1) English  <a href="#">View in Reaxys</a>  <b>Patent; Merrell Pharmaceuticals Inc.;</b> US5840729; (1998); (A1) English  <a href="#">View in Reaxys</a></p>